

Thermodynamic Stability and Anion Ordering of Perovskite Oxynitrides

Samuel D. Young, Jiadong Chen, Wenhao Sun, Bryan Goldsmith, Ghanshyam Pilia
AIChE National Meeting 2022 – 18 Nov 2022

Nitrogen chemistry can address nitrate fertilizer pollution

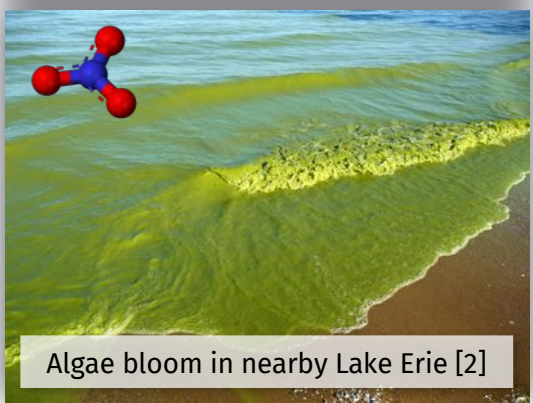
1. Fields, S. *Environmental Health Perspectives* **112**, A556–A563 (2004).
2. Erickson, J. Fishing in greener waters: Understanding the impact of harmful algal blooms on Lake Erie anglers. *University of Michigan News* (2018).
3. Wang, Z., Young, S. D., Goldsmith, B. R. & Singh, N. *Journal of Catalysis* **395**, 143–154 (2021).
4. Richards, D., Young, S. D., Goldsmith, B. R. & Singh, N. *Catal. Sci. Technol.* **11**, 7331–7346 (2021).



Nitrogen chemistry can address nitrate fertilizer pollution



Nitrate runoff from fertilizer [1]



Algae bloom in nearby Lake Erie [2]

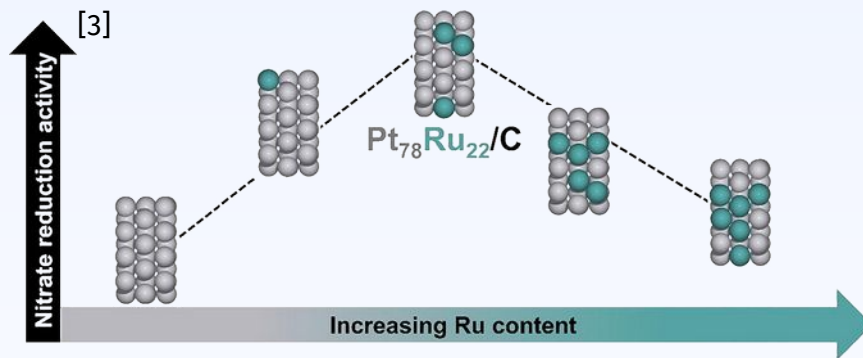
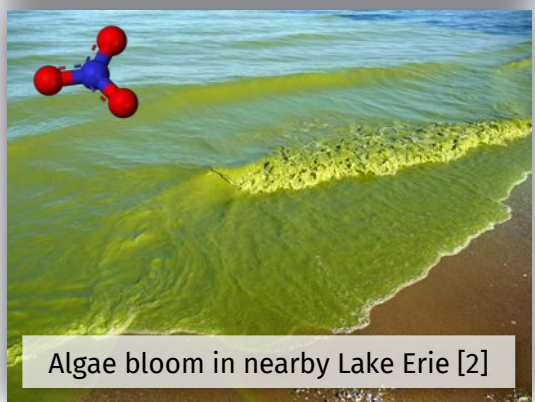
1. Fields, S. *Environmental Health Perspectives* **112**, A556–A563 (2004).

2. Erickson, J. Fishing in greener waters: Understanding the impact of harmful algal blooms on Lake Erie anglers. *University of Michigan News* (2018).

3. Wang, Z., Young, S. D., Goldsmith, B. R. & Singh, N. *Journal of Catalysis* **395**, 143–154 (2021).

4. Richards, D., Young, S. D., Goldsmith, B. R. & Singh, N. *Catal. Sci. Technol.* **11**, 7331–7346 (2021).

Nitrogen chemistry can address nitrate fertilizer pollution



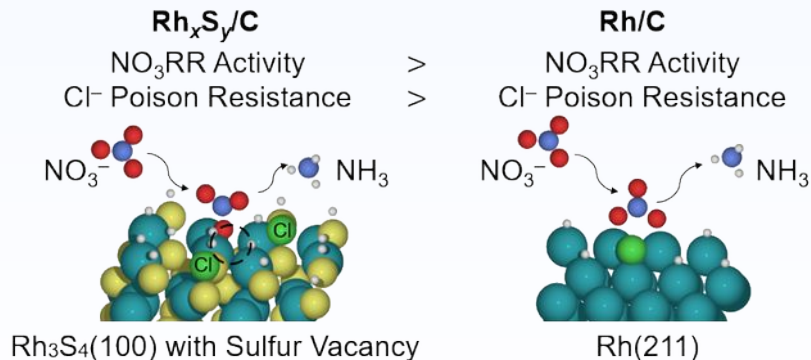
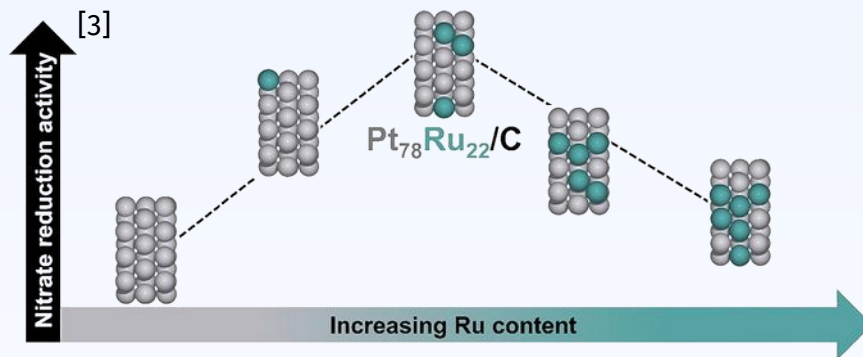
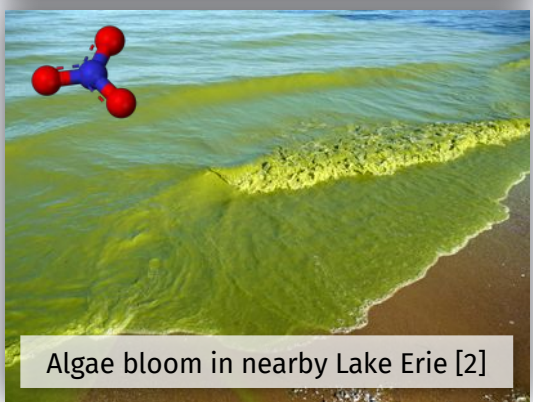
1. Fields, S. *Environmental Health Perspectives* **112**, A556–A563 (2004).

2. Erickson, J. Fishing in greener waters: Understanding the impact of harmful algal blooms on Lake Erie anglers. *University of Michigan News* (2018).

3. Wang, Z., Young, S. D., Goldsmith, B. R. & Singh, N. *Journal of Catalysis* **395**, 143–154 (2021).

4. Richards, D., Young, S. D., Goldsmith, B. R. & Singh, N. *Catal. Sci. Technol.* **11**, 7331–7346 (2021).

Nitrogen chemistry can address nitrate fertilizer pollution



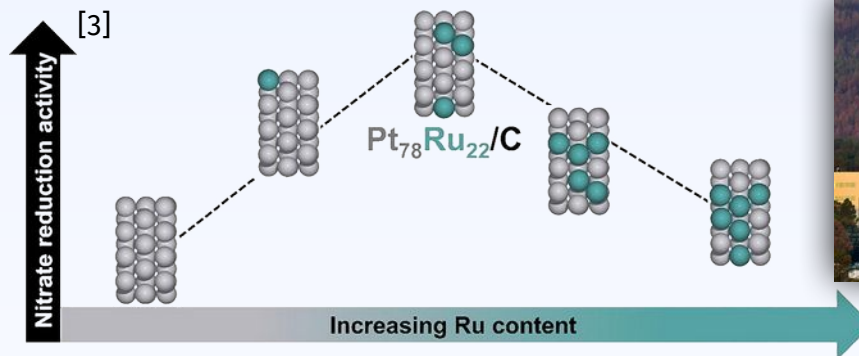
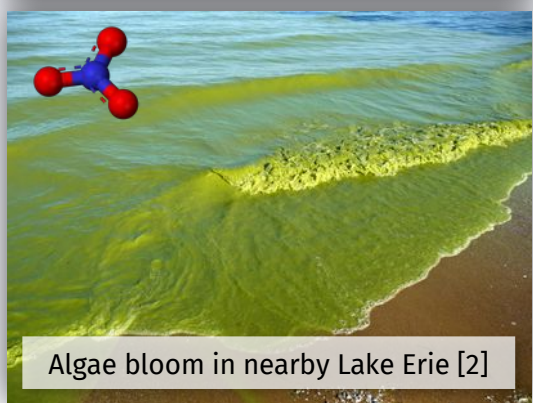
1. Fields, S. *Environmental Health Perspectives* **112**, A556–A563 (2004).

2. Erickson, J. Fishing in greener waters: Understanding the impact of harmful algal blooms on Lake Erie anglers. *University of Michigan News* (2018).

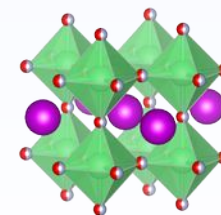
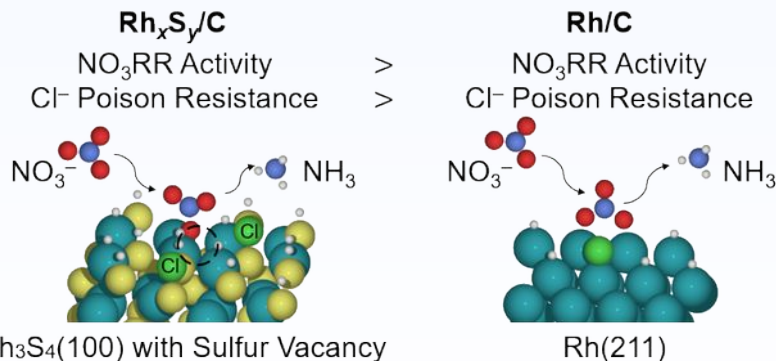
3. Wang, Z., Young, S. D., Goldsmith, B. R. & Singh, N. *Journal of Catalysis* **395**, 143–154 (2021).

4. Richards, D., Young, S. D., Goldsmith, B. R. & Singh, N. *Catal. Sci. Technol.* **11**, 7331–7346 (2021).

Nitrogen chemistry can address nitrate fertilizer pollution



Ghanshyam Pilania



1. Fields, S. *Environmental Health Perspectives* **112**, A556–A563 (2004).

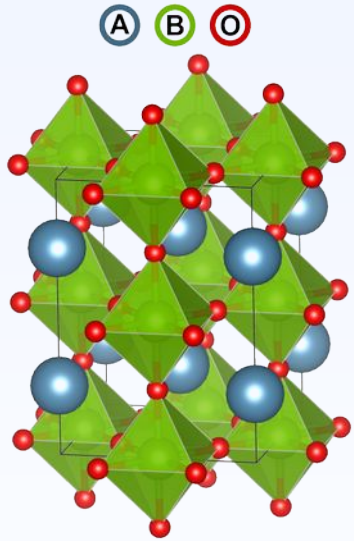
2. Erickson, J. Fishing in greener waters: Understanding the impact of harmful algal blooms on Lake Erie anglers. *University of Michigan News* (2018).

3. Wang, Z., Young, S. D., Goldsmith, B. R. & Singh, N. *Journal of Catalysis* **395**, 143–154 (2021).

4. Richards, D., Young, S. D., Goldsmith, B. R. & Singh, N. *Catal. Sci. Technol.* **11**, 7331–7346 (2021).

Perovskite oxynitrides (PONs) are important for many applications

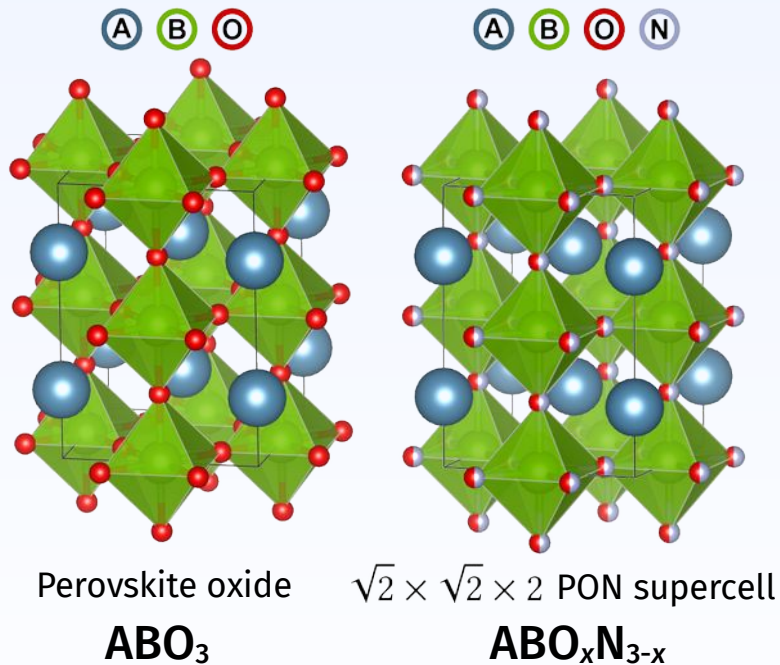
Perovskite oxynitrides (PONs) are important for many applications



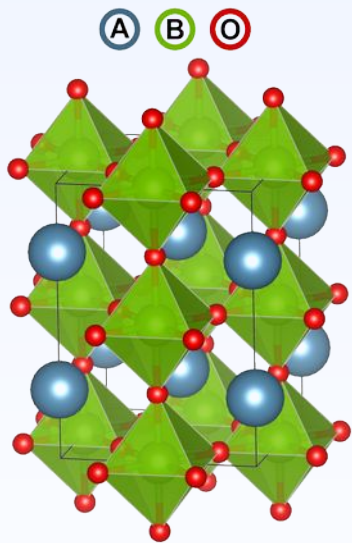
Perovskite oxide



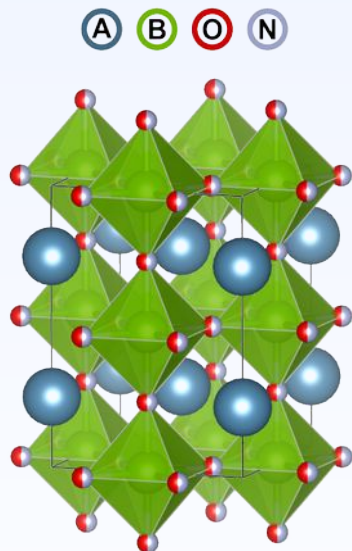
Perovskite oxynitrides (PONs) are important for many applications



Perovskite oxynitrides (PONs) are important for many applications



Perovskite oxide
ABO₃



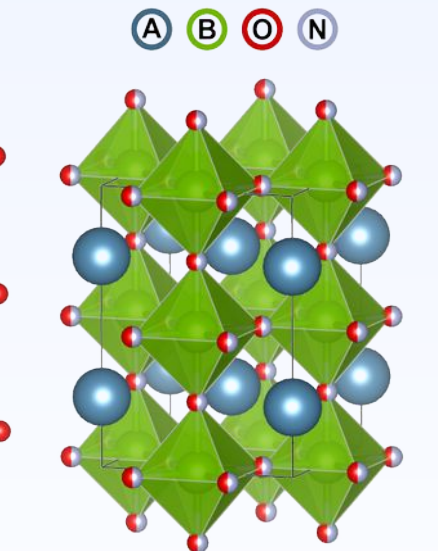
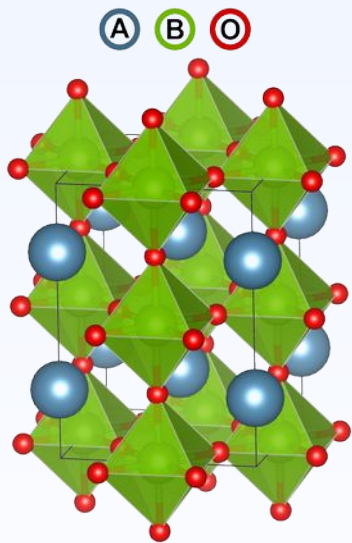
$\sqrt{2} \times \sqrt{2} \times 2$ PON supercell
ABO_xN_{3-x}

Pigments



Sakata, T., et al. *Inorg. Chem.* **60**, 7, 4852–4859 (2021)

Perovskite oxynitrides (PONs) are important for many applications

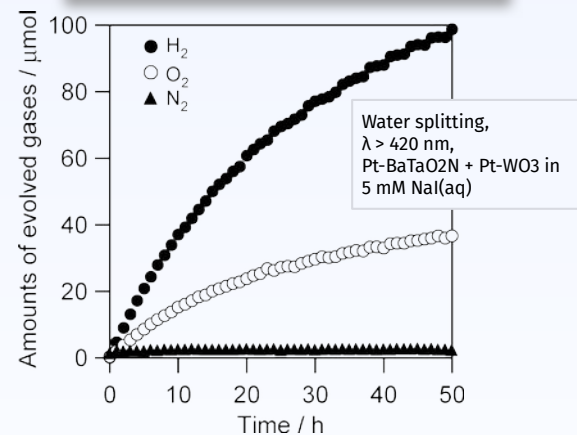


Pigments



Sakata, T., et al. *Inorg. Chem.* **60**, 7, 4852–4859 (2021)

Photocatalysis



Higashi, M., et al. *Chem. Mater.* **21**, 1543–1549 (2009).

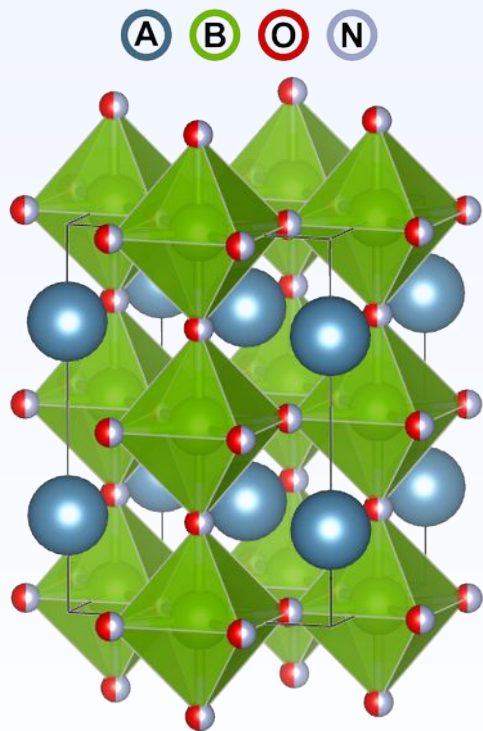
The structure and composition of a PON strongly impacts its performance and stability.

Fuertes, A. Chemistry and applications of oxynitride perovskites. *J. Mater. Chem.* **22**, 3293–3299 (2012).

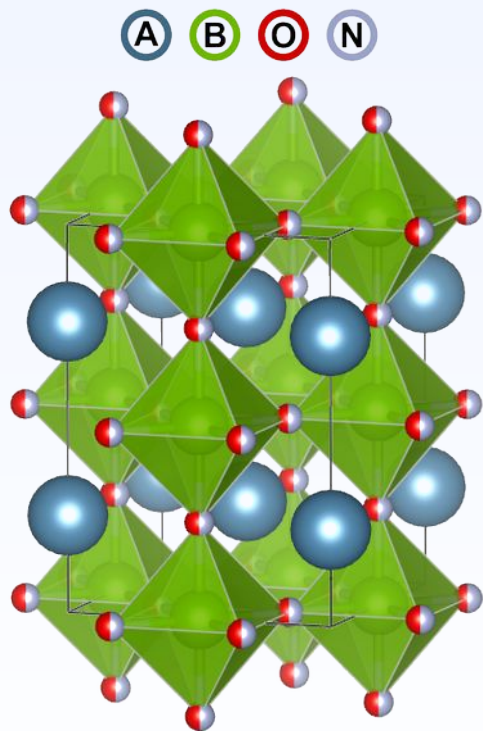
There are many choices to make when designing a stable PON



There are many choices to make when designing a stable PON

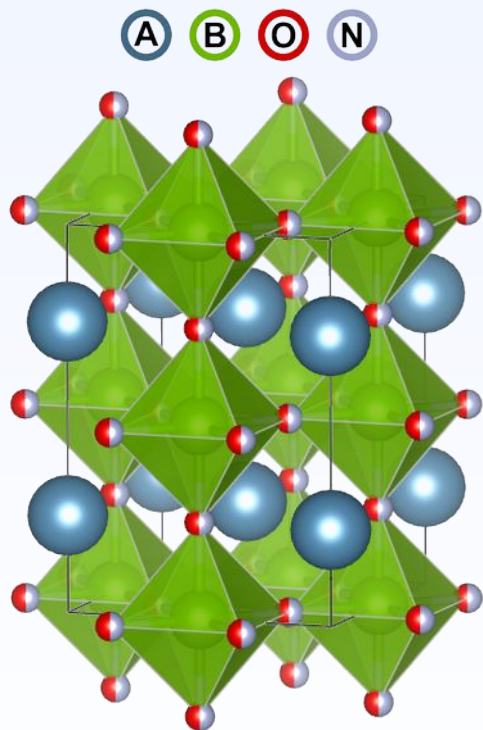


There are many choices to make when designing a stable PON



How should we choose the chemistry and structure to maximize stability?

There are many choices to make when designing a stable PON



How should we choose the chemistry and structure to maximize stability?

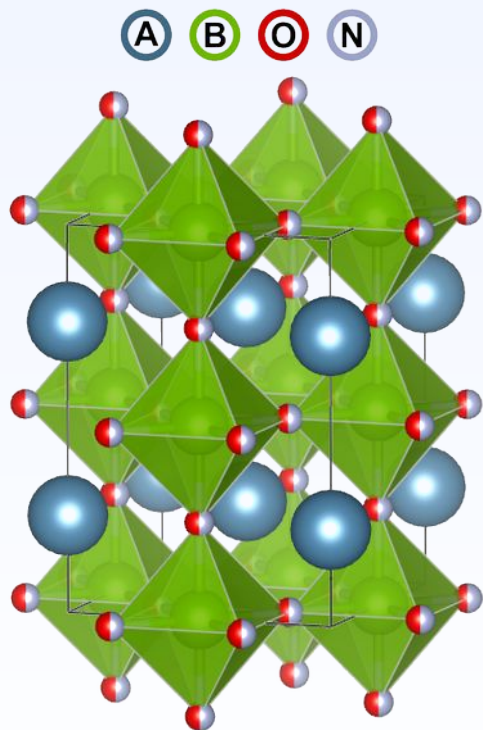


A Ca Ca La La ... ?

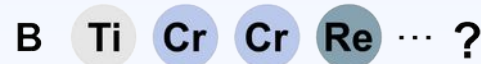
B Ti Cr Cr Re ... ?

Which cation pairs?

There are many choices to make when designing a stable PON

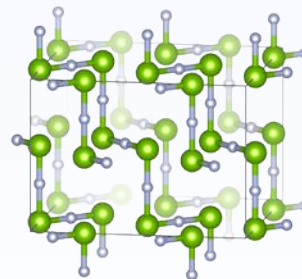


How should we choose the chemistry and structure to maximize stability?

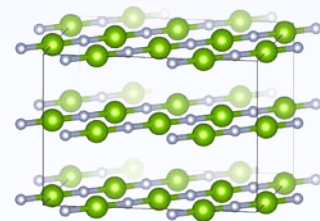


Which cation pairs?

Mostly *cis*



Mostly *trans*



or

or ?

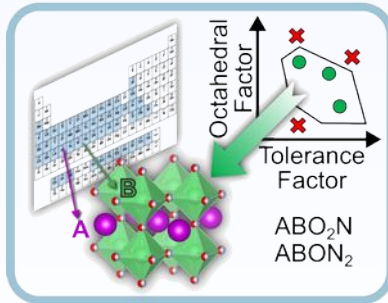
Which anion orderings? Trends?

Goal: determine thermodynamic stability and anion ordering in ABO_2N and $ABON_2$ perovskite oxynitrides

Goal: determine thermodynamic stability and anion ordering in ABO_2N and $ABON_2$ perovskite oxynitrides

(a) Cation Pair Selection

- Select elements from periodic table
- Enumerate all permutations of cations
- Filter based on geometric factors



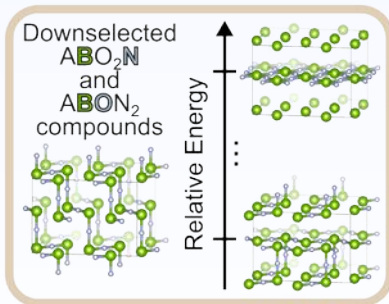
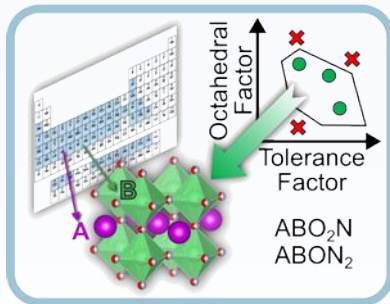
Goal: determine thermodynamic stability and anion ordering in ABO_2N and $ABON_2$ perovskite oxynitrides

(a) Cation Pair Selection

- Select elements from periodic table
- Enumerate all permutations of cations
- Filter based on geometric factors

(b) Anion Ordering Selection

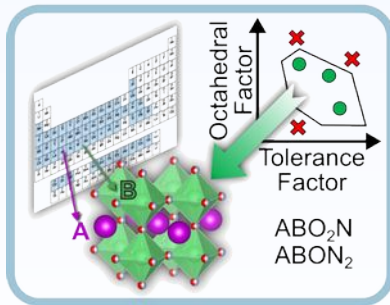
- Enumerate distinct anion orderings with selected cation pairs
- Evaluate relative energies of anion orderings



Goal: determine thermodynamic stability and anion ordering in ABO_2N and $ABON_2$ perovskite oxynitrides

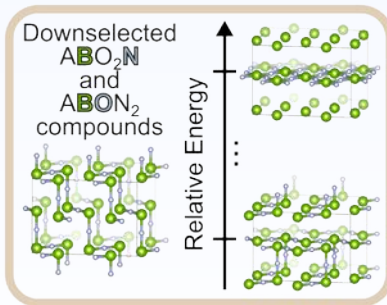
(a) Cation Pair Selection

- Select elements from periodic table
- Enumerate all permutations of cations
- Filter based on geometric factors



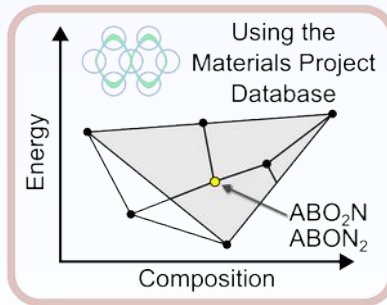
(b) Anion Ordering Selection

- Enumerate distinct anion orderings with selected cation pairs
- Evaluate relative energies of anion orderings



(c) Energy-Above-Hull Analysis

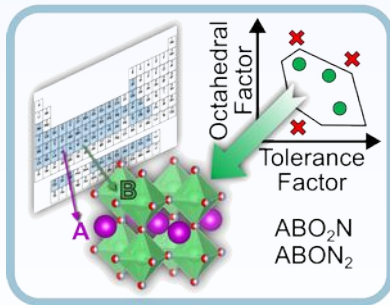
- Pair 295 cation pairs with optimal anion ordering
- Calculate energies above hull using Materials Project



Goal: determine thermodynamic stability and anion ordering in ABO_2N and $ABON_2$ perovskite oxynitrides

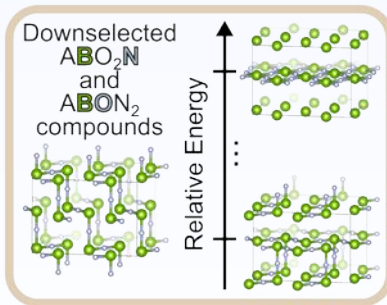
(a) Cation Pair Selection

- Select elements from periodic table
- Enumerate all permutations of cations
- Filter based on geometric factors



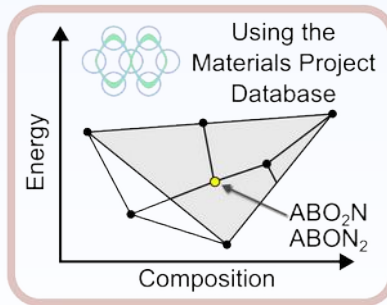
(b) Anion Ordering Selection

- Enumerate distinct anion orderings with selected cation pairs
- Evaluate relative energies of anion orderings



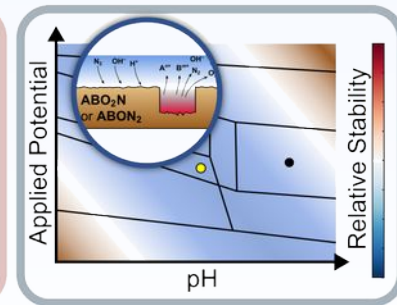
(c) Energy-Above-Hull Analysis

- Pair 295 cation pairs with optimal anion ordering
- Calculate energies above hull using Materials Project



(d) Electrochemical Stability Analysis

- Construct multidimensional Pourbaix diagrams
- Identify regions of stability and corresponding operating conditions



Select cation pairs

- We build an experimental stability hull from known stable PONs.^[1, 2]

1. Li, W., Ionescu, E., Riedel, R. & Gurlo, A. *J. Mater. Chem. A* **1**, 12239 (2013).
2. Wang, H.-C., Schmidt, J., Botti, S. & L. Marques, M. A. *J. Mater. Chem. A* **9**, 8501–8513 (2021).
3. Higashi, M., *et al.* *Chem. Mater.* **21**, 1543–1549 (2009).



Select cation pairs

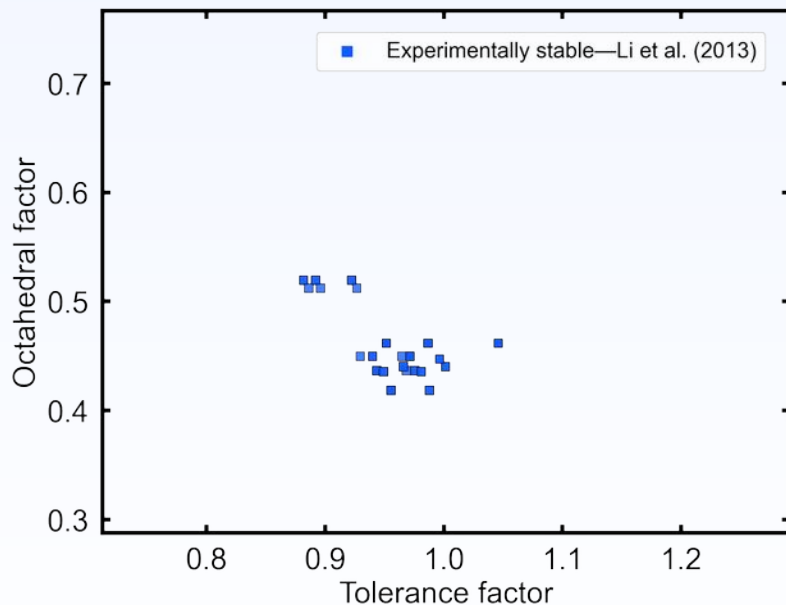
- We build an experimental stability hull from known stable PONs.^[1, 2]

Stoichiometry	Goldschmidt tolerance factor	Octahedral factor
ABO ₂ N	$\frac{[(r_A + r_O)^8(r_A + r_N)^4]^{1/12}}{\sqrt{2}[(r_B + r_O)^4(r_B + r_N)^2]^{1/6}}$	$\frac{r_B}{(r_O^4 r_N^2)^{1/6}}$
ABON ₂	$\frac{[(r_A + r_O)^4(r_A + r_N)^8]^{1/12}}{\sqrt{2}[(r_B + r_O)^2(r_B + r_N)^4]^{1/6}}$	$\frac{r_B}{(r_O^2 r_N^4)^{1/6}}$

- Li, W., Ionescu, E., Riedel, R. & Gurlo, A. *J. Mater. Chem. A* **1**, 12239 (2013).
- Wang, H.-C., Schmidt, J., Botti, S. & L. Marques, M. A. *J. Mater. Chem. A* **9**, 8501–8513 (2021).
- Higashi, M., *et al.* *Chem. Mater.* **21**, 1543–1549 (2009).

Select cation pairs

- We build an experimental stability hull from known stable PONs.^[1, 2]

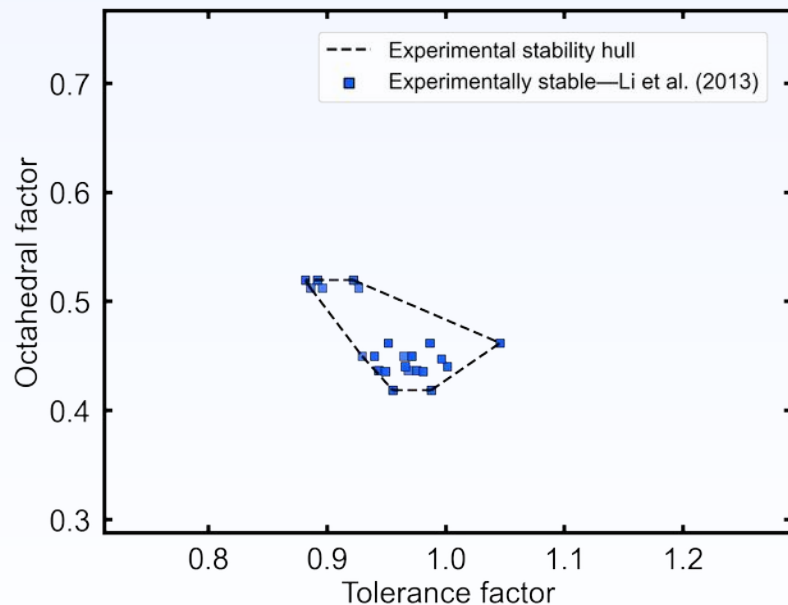


Stoichiometry	Goldschmidt tolerance factor	Octahedral factor
ABO_2N	$\frac{[(r_A + r_O)^8(r_A + r_N)^4]^{1/12}}{\sqrt{2}[(r_B + r_O)^4(r_B + r_N)^2]^{1/6}}$	$\frac{r_B}{(r_O^4 r_N^2)^{1/6}}$
ABON_2	$\frac{[(r_A + r_O)^4(r_A + r_N)^8]^{1/12}}{\sqrt{2}[(r_B + r_O)^2(r_B + r_N)^4]^{1/6}}$	$\frac{r_B}{(r_O^2 r_N^4)^{1/6}}$

- Li, W., Ionescu, E., Riedel, R. & Gurlo, A. *J. Mater. Chem. A* **1**, 12239 (2013).
- Wang, H.-C., Schmidt, J., Botti, S. & L. Marques, M. A. *J. Mater. Chem. A* **9**, 8501–8513 (2021).
- Higashi, M., et al. *Chem. Mater.* **21**, 1543–1549 (2009).

Select cation pairs

- We build an experimental stability hull from known stable PONs.^[1, 2]

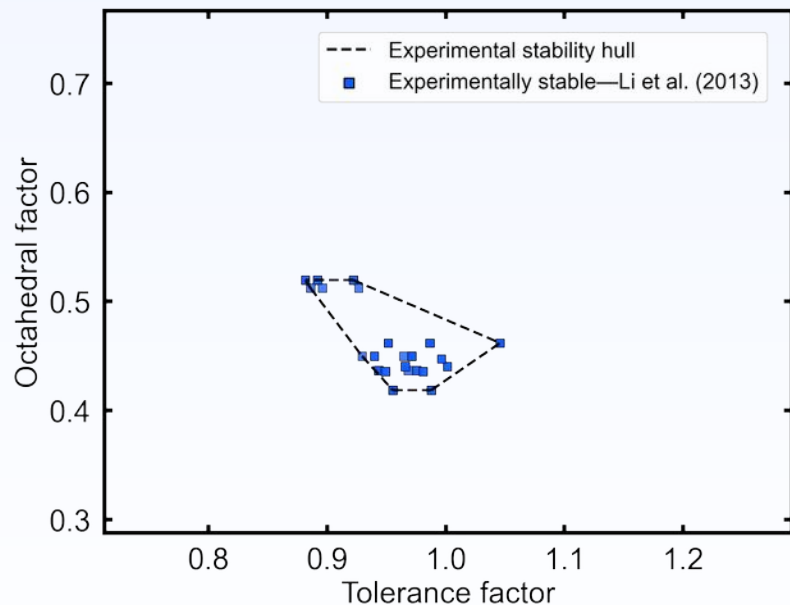


Stoichiometry	Goldschmidt tolerance factor	Octahedral factor
ABO_2N	$\frac{[(r_A + r_O)^8(r_A + r_N)^4]^{1/12}}{\sqrt{2}[(r_B + r_O)^4(r_B + r_N)^2]^{1/6}}$	$\frac{r_B}{(r_O^4 r_N^2)^{1/6}}$
ABON_2	$\frac{[(r_A + r_O)^4(r_A + r_N)^8]^{1/12}}{\sqrt{2}[(r_B + r_O)^2(r_B + r_N)^4]^{1/6}}$	$\frac{r_B}{(r_O^2 r_N^4)^{1/6}}$

- Li, W., Ionescu, E., Riedel, R. & Gurlo, A. *J. Mater. Chem.* **A** 1, 12239 (2013).
- Wang, H.-C., Schmidt, J., Botti, S. & L. Marques, M. A. *J. Mater. Chem. A* **9**, 8501–8513 (2021).
- Higashi, M., et al. *Chem. Mater.* **21**, 1543–1549 (2009).

Select cation pairs

- We build an experimental stability hull from known stable PONs.^[1, 2]



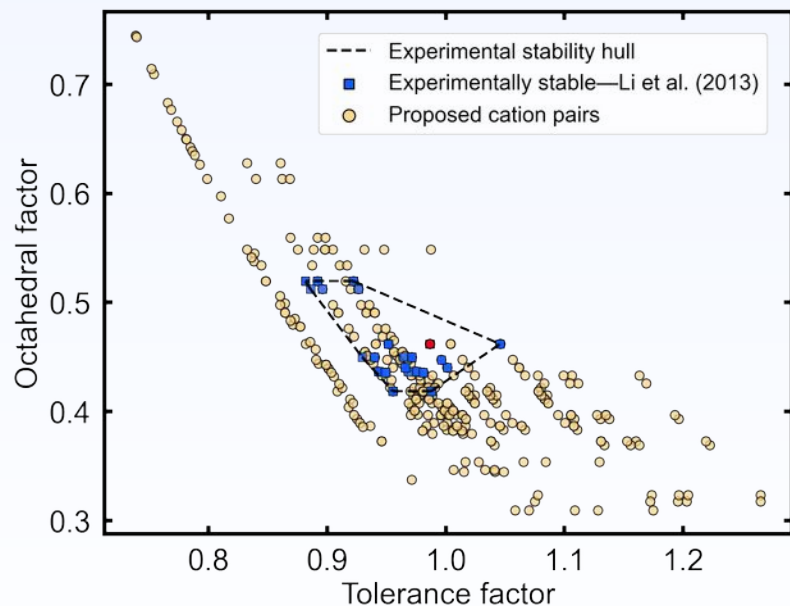
Stoichiometry	Goldschmidt tolerance factor	Octahedral factor
ABO_2N	$\frac{[(r_A + r_O)^8(r_A + r_N)^4]^{1/12}}{\sqrt{2}[(r_B + r_O)^4(r_B + r_N)^2]^{1/6}}$	$\frac{r_B}{(r_O^4 r_N^2)^{1/6}}$
$ABON_2$	$\frac{[(r_A + r_O)^4(r_A + r_N)^8]^{1/12}}{\sqrt{2}[(r_B + r_O)^2(r_B + r_N)^4]^{1/6}}$	$\frac{r_B}{(r_O^2 r_N^4)^{1/6}}$

1 H																	2 He
3 Li	4 Be											5 B	6 C	7 N	8 O	9 F	10 Ne
11 Na	12 Mg											13 Al	14 Si	15 P	16 S	17 Cl	18 Ar
19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr
37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe
55 Cs	56 Ba	57 La	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn
87 Fr	88 Ra	89 Ac	104 Rf	105 Db	106 Sg	107 Bh	108 Hs	109 Mt	110 Ds	111 Rg	112 Cn	113 Nh	114 Fl	115 Mc	116 Lv	117 Ts	118 Og
58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu				
90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No	103 Lr				

- Li, W., Ionescu, E., Riedel, R. & Gurlo, A. *J. Mater. Chem. A* **1**, 12239 (2013).
- Wang, H.-C., Schmidt, J., Botti, S. & L. Marques, M. A. *J. Mater. Chem. A* **9**, 8501–8513 (2021).
- Higashi, M., et al. *Chem. Mater.* **21**, 1543–1549 (2009).

Select cation pairs

- We build an experimental stability hull from known stable PONs.^[1, 2]



Stoichiometry	Goldschmidt tolerance factor	Octahedral factor
ABO_2N	$\frac{[(r_A + r_O)^8(r_A + r_N)^4]^{1/12}}{\sqrt{2}[(r_B + r_O)^4(r_B + r_N)^2]^{1/6}}$	$\frac{r_B}{(r_O^4 r_N^2)^{1/6}}$
$ABON_2$	$\frac{[(r_A + r_O)^4(r_A + r_N)^8]^{1/12}}{\sqrt{2}[(r_B + r_O)^2(r_B + r_N)^4]^{1/6}}$	$\frac{r_B}{(r_O^2 r_N^4)^{1/6}}$

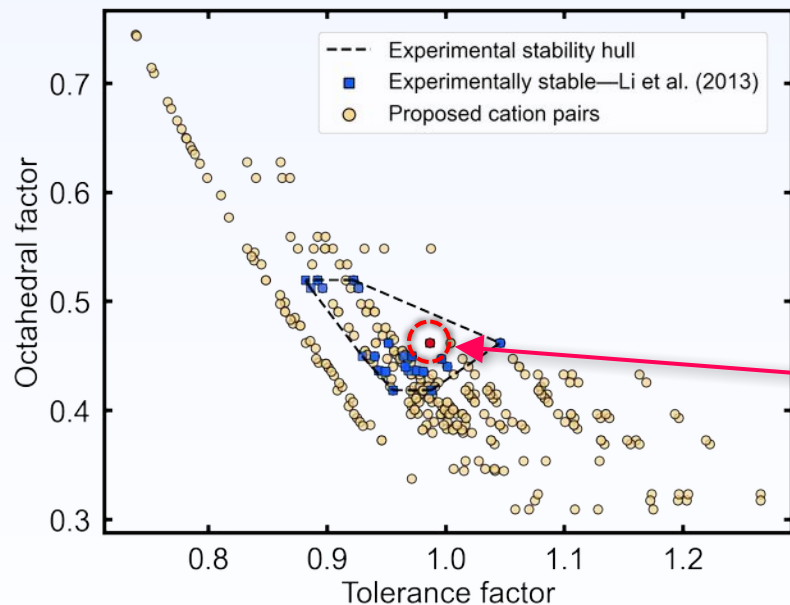
1 H																	2 He
3 Li	4 Be											5 B	6 C	7 N	8 O	9 F	10 Ne
11 Na	12 Mg											13 Al	14 Si	15 P	16 S	17 Cl	18 Ar
19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr
37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe
55 Cs	56 Ba	57 La	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn
87 Fr	88 Ra	89 Ac	104 Rf	105 Db	106 Sg	107 Bh	108 Hs	109 Mt	110 Ds	111 Rg	112 Cn	113 Nh	114 Fl	115 Mc	116 Lv	117 Ts	118 Og
58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu				
90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No	103 Lr				

316 enumerated compounds
101 (32%) inside hull; 215 (68%) outside hull

- Li, W., Ionescu, E., Riedel, R. & Gurlo, A. *J. Mater. Chem. A* **1**, 12239 (2013).
- Wang, H.-C., Schmidt, J., Botti, S. & L. Marques, M. A. *J. Mater. Chem. A* **9**, 8501–8513 (2021).
- Higashi, M., et al. *Chem. Mater.* **21**, 1543–1549 (2009).

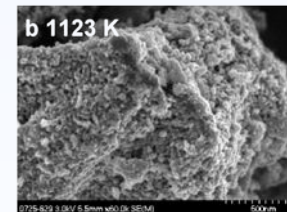
Select cation pairs

- We build an experimental stability hull from known stable PONs.^[1, 2]



Stoichiometry	Goldschmidt tolerance factor	Octahedral factor
ABO_2N	$\frac{[(r_A + r_O)^8(r_A + r_N)^4]^{1/12}}{\sqrt{2}[(r_B + r_O)^4(r_B + r_N)^2]^{1/6}}$	$\frac{r_B}{(r_O^4 r_N^2)^{1/6}}$
$ABON_2$	$\frac{[(r_A + r_O)^4(r_A + r_N)^8]^{1/12}}{\sqrt{2}[(r_B + r_O)^2(r_B + r_N)^4]^{1/6}}$	$\frac{r_B}{(r_O^2 r_N^4)^{1/6}}$

1 H																	2 He																												
3 Li	4 Be											5 B	6 C	7 N	8 O	9 F	10 Ne																												
11 Na	12 Mg											13 Al	14 Si	15 P	16 S	17 Cl	18 Ar																												
19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr																												
37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe																												
55 Cs	56 Ba	57 La	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn																												
87 Fr	88 Ra	89 Ac	104 Rf	105 Db	106 Sg	107 Bh	108 Hs	109 Mt	110 Ds	111 Rg	112 Cn	113 Nh	114 Fl	115 Mc	116 Lv	117 Ts	118 Og																												
<table border="1"> <tbody> <tr> <td>58 Ce</td> <td>59 Pr</td> <td>60 Nd</td> <td>61 Pm</td> <td>62 Sm</td> <td>63 Eu</td> <td>64 Gd</td> <td>65 Tb</td> <td>66 Dy</td> <td>67 Ho</td> <td>68 Er</td> <td>69 Tm</td> <td>70 Yb</td> <td>71 Lu</td> </tr> <tr> <td>90 Th</td> <td>91 Pa</td> <td>92 U</td> <td>93 Np</td> <td>94 Pu</td> <td>95 Am</td> <td>96 Cm</td> <td>97 Bk</td> <td>98 Cf</td> <td>99 Es</td> <td>100 Fm</td> <td>101 Md</td> <td>102 No</td> <td>103 Lr</td> </tr> </tbody> </table>																		58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu	90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No	103 Lr
58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu																																
90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No	103 Lr																																



$SrNbO_2N$ [3]

316 enumerated compounds
 101 (32%) inside hull; 215 (68%) outside hull

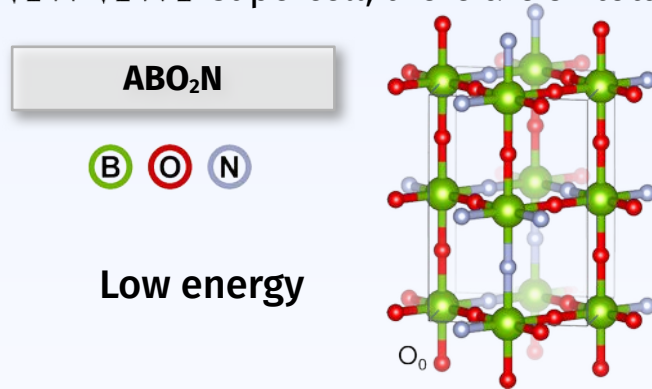
- Li, W., Ionescu, E., Riedel, R. & Gurlo, A. *J. Mater. Chem. A* **1**, 12239 (2013).
- Wang, H.-C., Schmidt, J., Botti, S. & L. Marques, M. A. *J. Mater. Chem. A* **9**, 8501–8513 (2021).
- Higashi, M., et al. *Chem. Mater.* **21**, 1543–1549 (2009).

We aim to identify preferred anion orderings

- For $\sqrt{2} \times \sqrt{2} \times 2$ supercell, there are 32 total symmetrically distinct anion orderings.^[1]

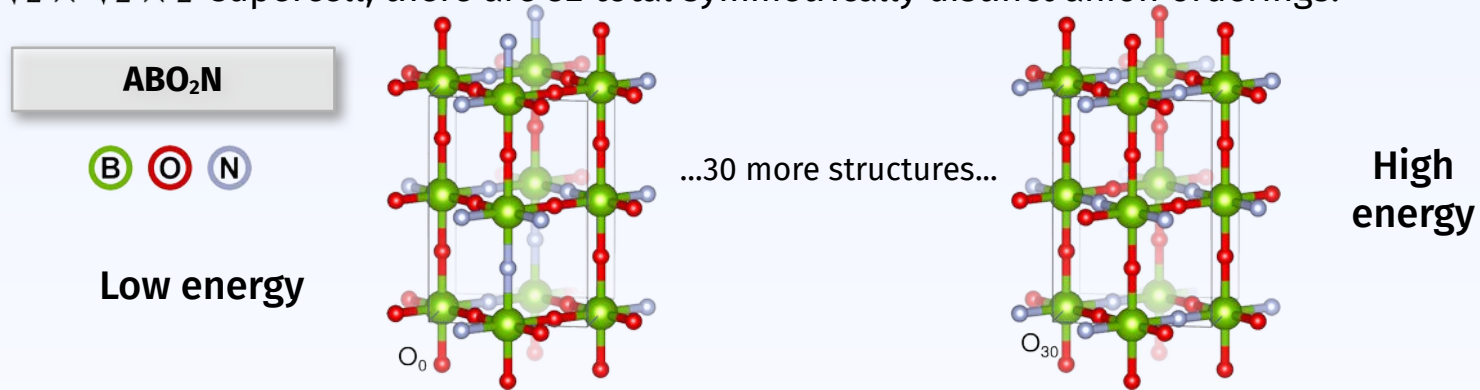
We aim to identify preferred anion orderings

- For $\sqrt{2} \times \sqrt{2} \times 2$ supercell, there are 32 total symmetrically distinct anion orderings.^[1]



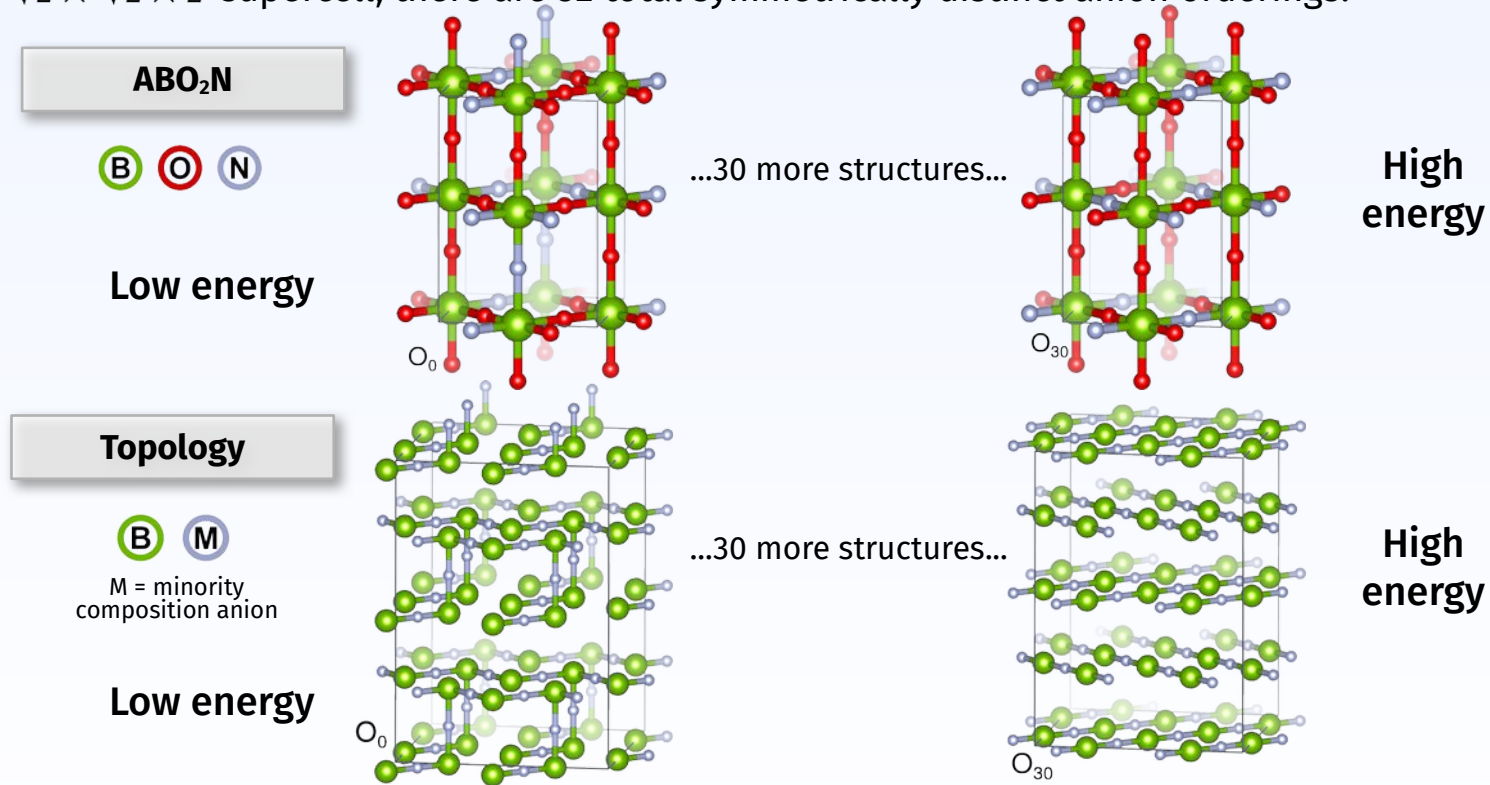
We aim to identify preferred anion orderings

- For $\sqrt{2} \times \sqrt{2} \times 2$ supercell, there are 32 total symmetrically distinct anion orderings.^[1]



We aim to identify preferred anion orderings

- For $\sqrt{2} \times \sqrt{2} \times 2$ supercell, there are 32 total symmetrically distinct anion orderings.^[1]



1. Hart, G. L. W., Nelson, L. J. & Forcade, R. W. Generating derivative structures at a fixed concentration. *Computational Materials Science* **59**, 101–107 (2012).

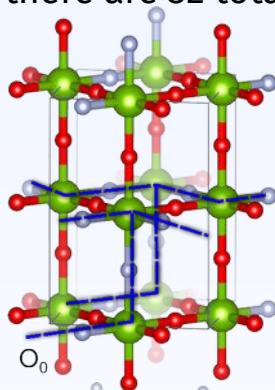
We aim to identify preferred anion orderings

- For $\sqrt{2} \times \sqrt{2} \times 2$ supercell, there are 32 total symmetrically distinct anion orderings.^[1]

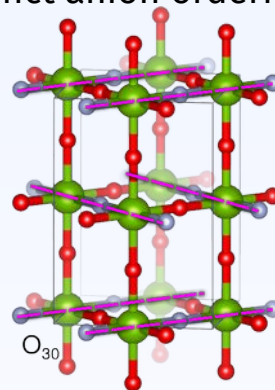
ABO₂N



Low energy

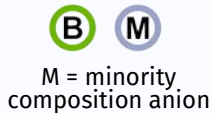


...30 more structures...

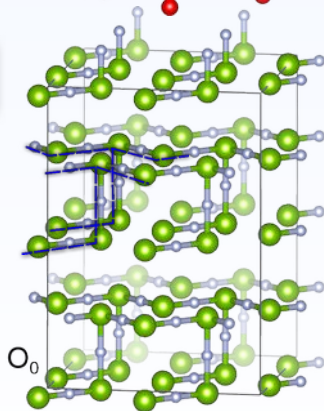


High energy

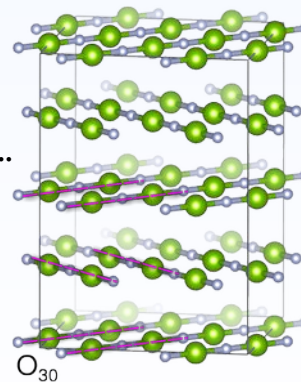
Topology



Low energy



...30 more structures...



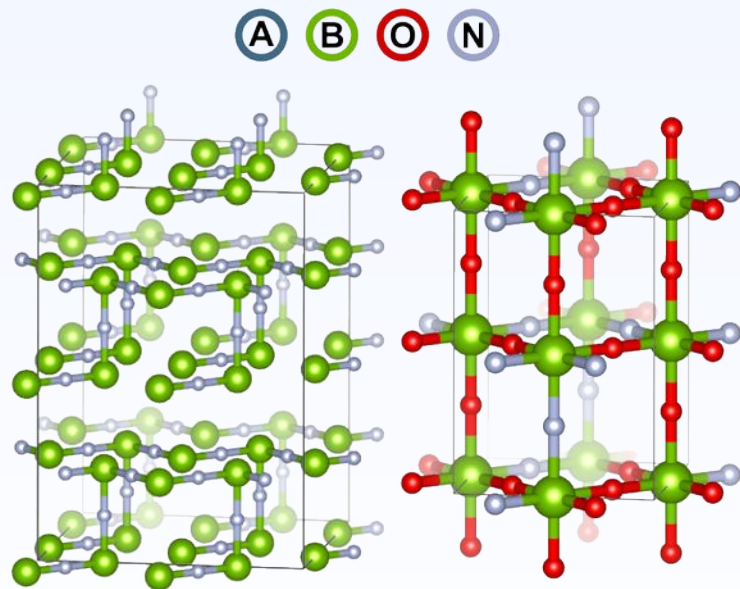
High energy

1. Hart, G. L. W., Nelson, L. J. & Forcade, R. W. Generating derivative structures at a fixed concentration. *Computational Materials Science* **59**, 101–107 (2012).

One anion ordering is consistently stable across 16 cation pairs

One anion ordering is consistently stable across 16 cation pairs

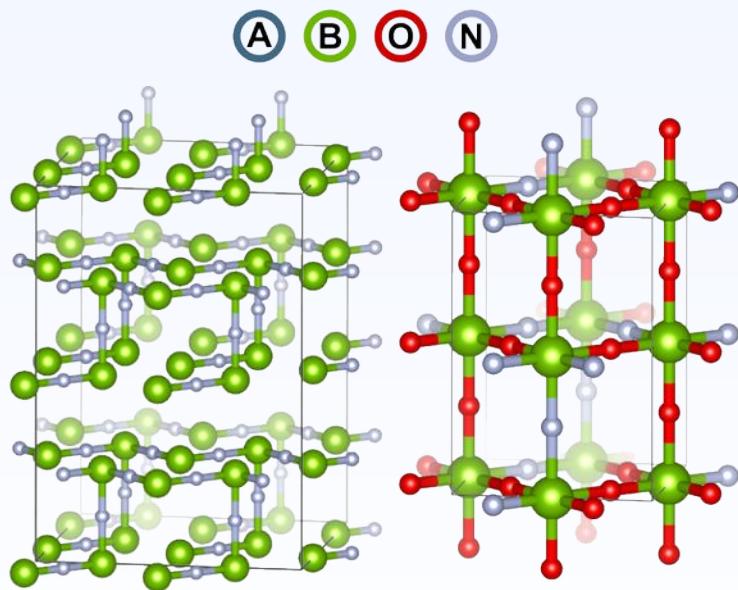
The most stable anion ordering contains
cis bonding across B atoms.



Ordering O_0

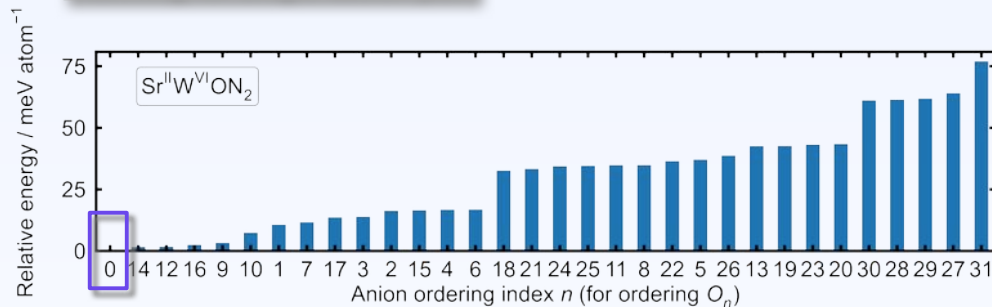
One anion ordering is consistently stable across 16 cation pairs

The most stable anion ordering contains cis bonding across B atoms.



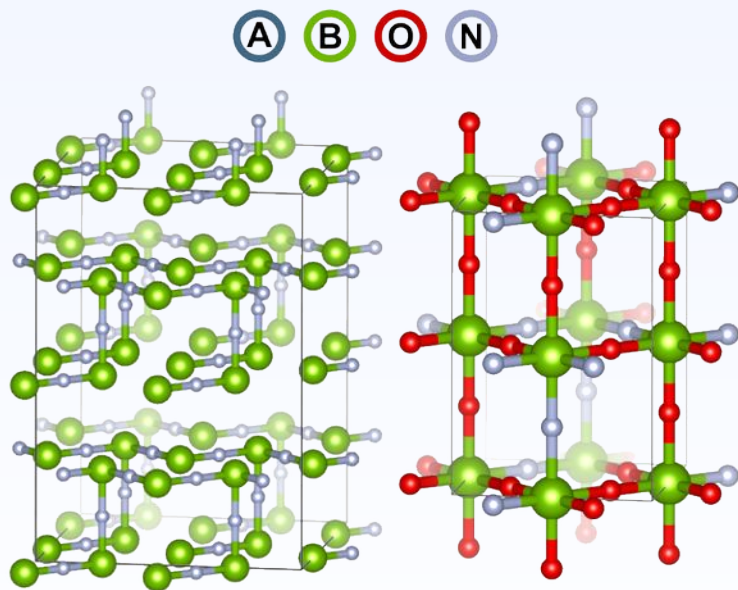
Ordering O_0

Ranking for single cation pair



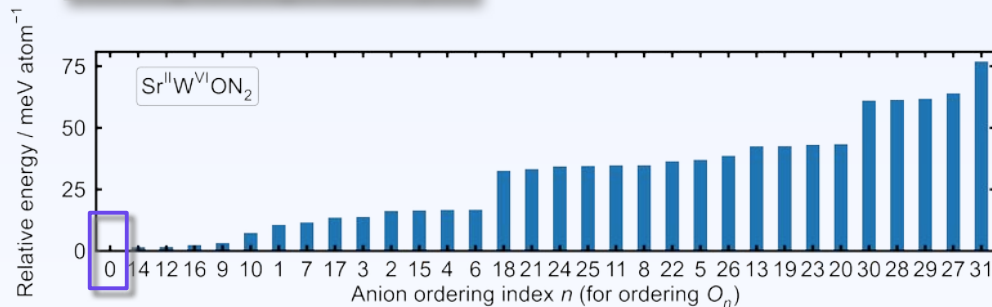
One anion ordering is consistently stable across 16 cation pairs

The most stable anion ordering contains cis bonding across B atoms.

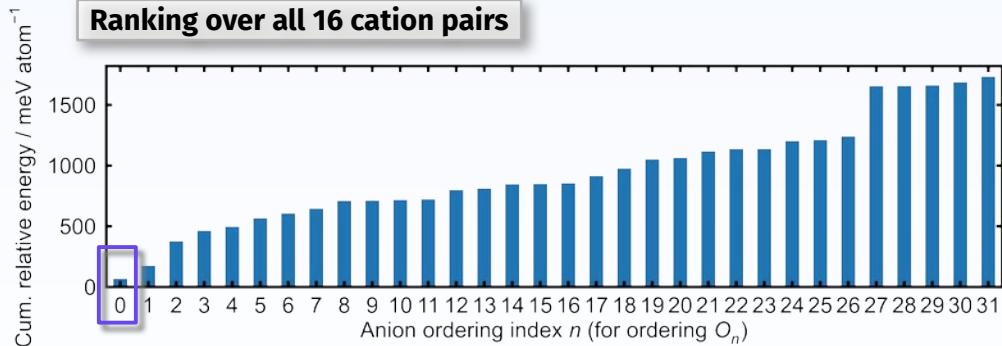


Ordering O_0

Ranking for single cation pair



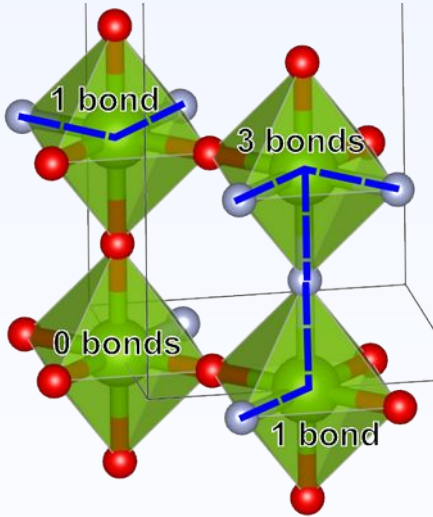
Ranking over all 16 cation pairs



Low-energy orderings have a high degree of *cis* bonding

Low-energy orderings have a high degree of *cis* bonding

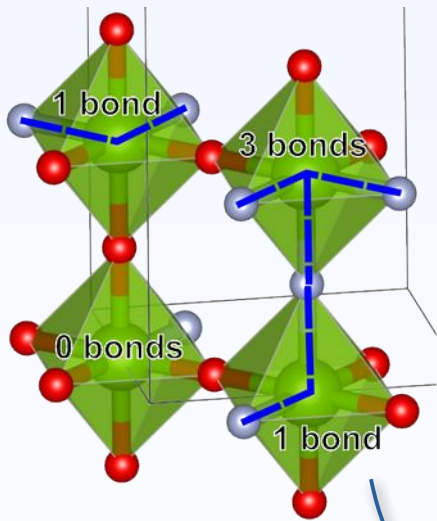
Count the M—B—M bonds



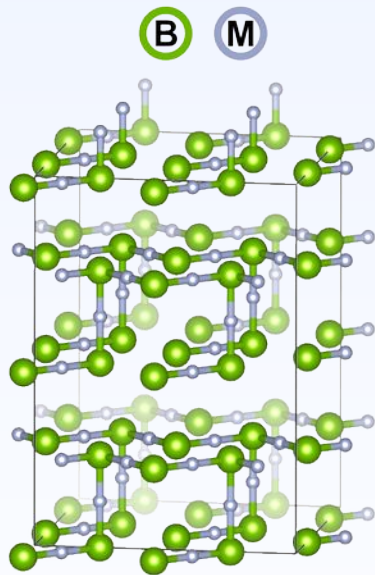
O_0 unit cell
All *cis* bonds

Low-energy orderings have a high degree of *cis* bonding

Count the M—B—M bonds



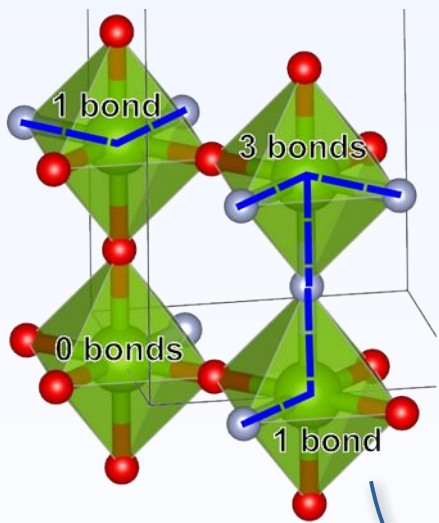
O_0 unit cell
All *cis* bonds



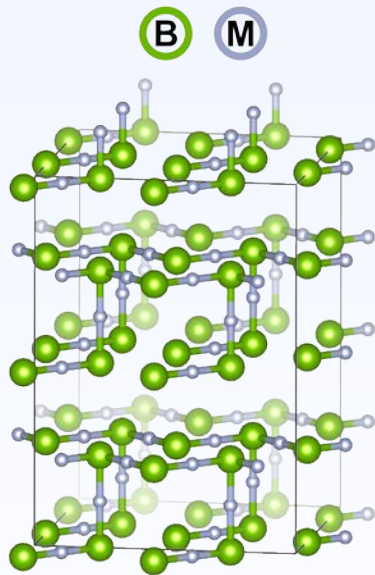
O_0
100% global *cis* bonding
Cis counts: 1, 0, 3, 1
3/4 octahedra with *cis* bonds

Low-energy orderings have a high degree of *cis* bonding

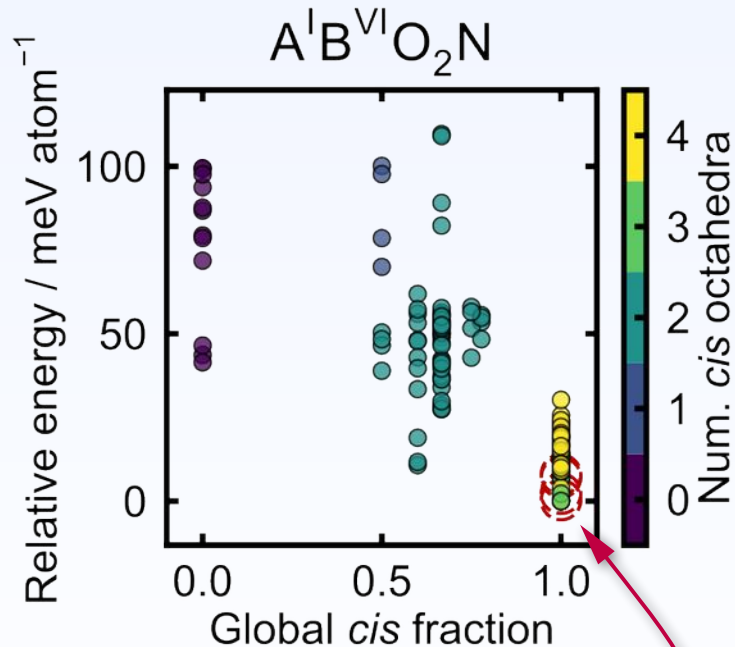
Count the M—B—M bonds



O_0 unit cell
All *cis* bonds



O_0
100% global *cis* bonding
Cis counts: 1, 0, 3, 1
3/4 octahedra with *cis* bonds



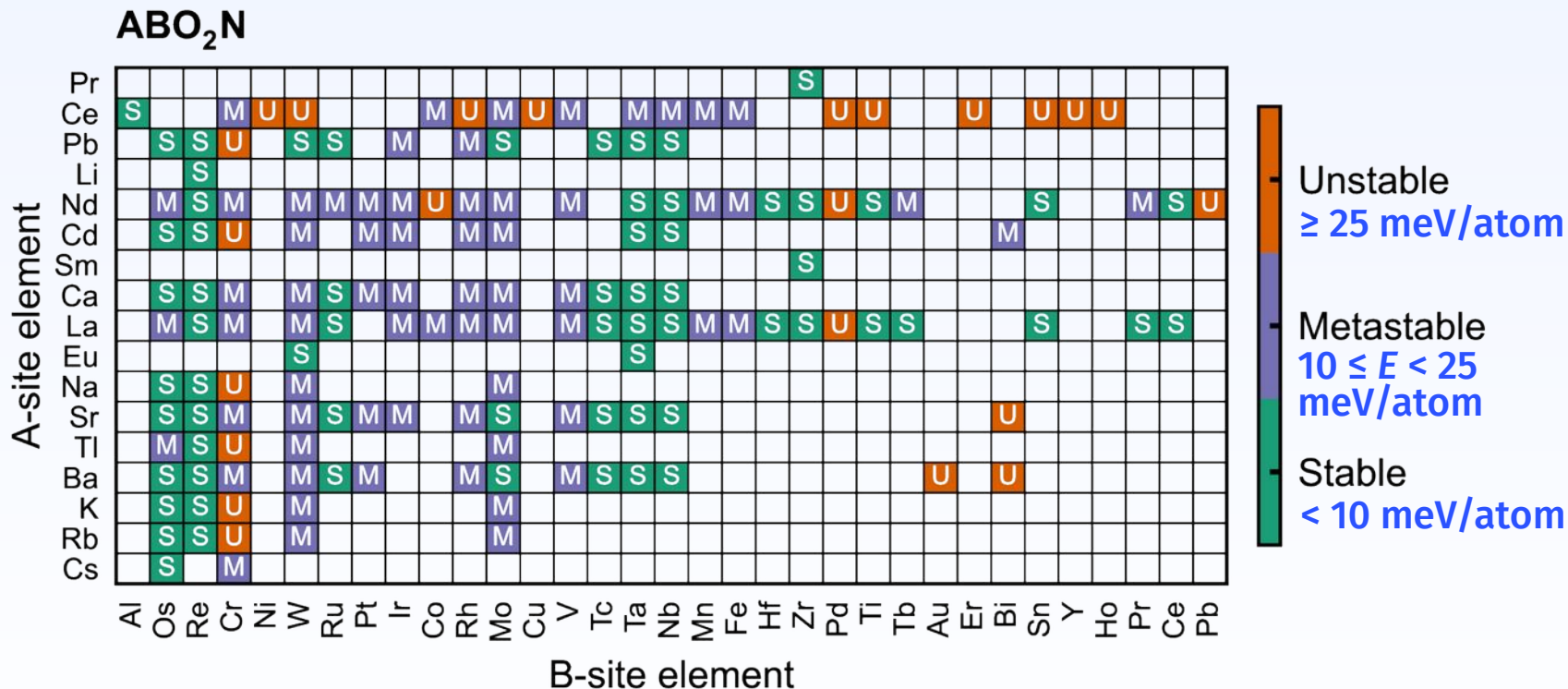
O_0 always has 100% global *cis* fraction

We identify 66 stable ABO_2N PON materials

1. Wang, H.-C., Schmidt, J., Botti, S. & L. Marques, M. A. A high-throughput study of oxynitride, oxyfluoride and nitrofluoride perovskites. *Journal of Materials Chemistry A* **9**, 8501–8513 (2021).

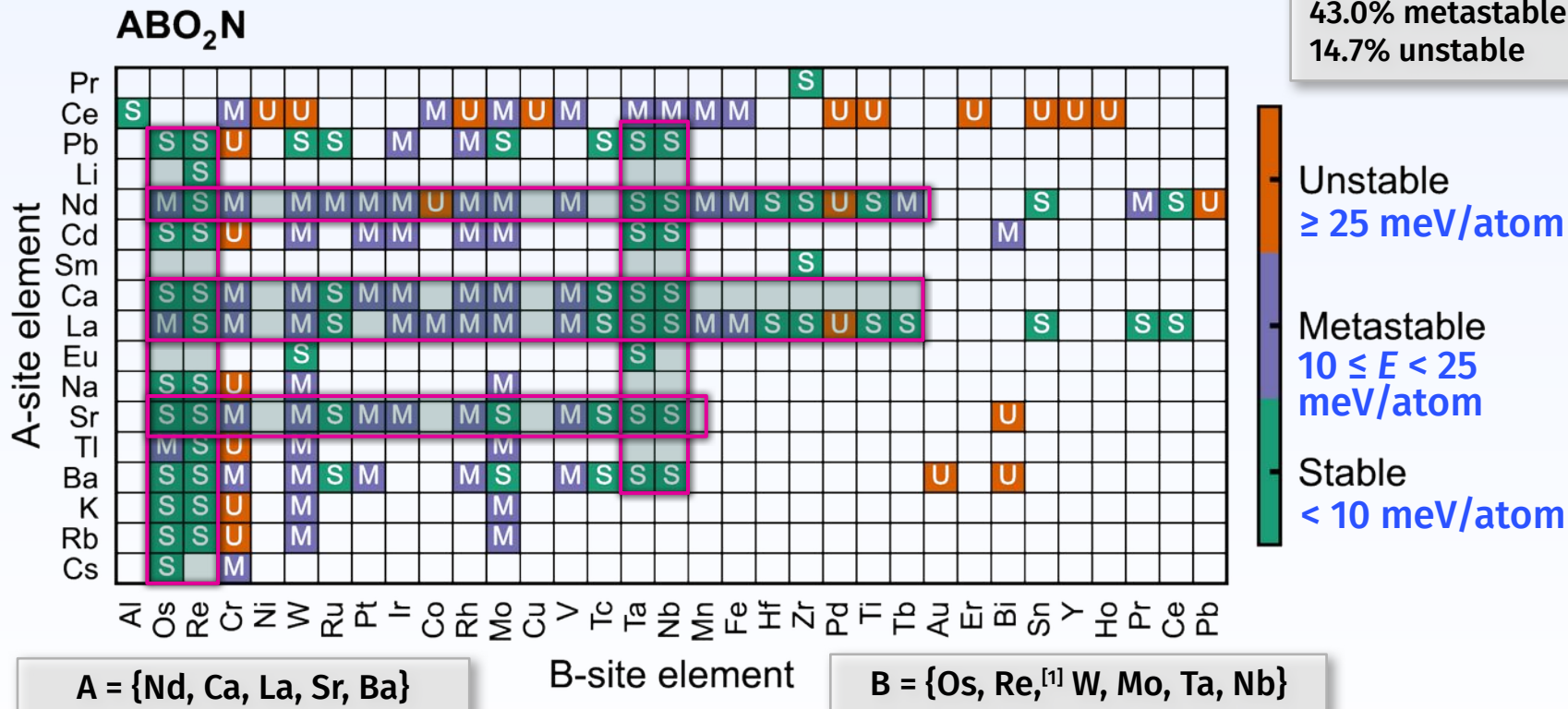


We identify 66 stable ABO₂N PON materials



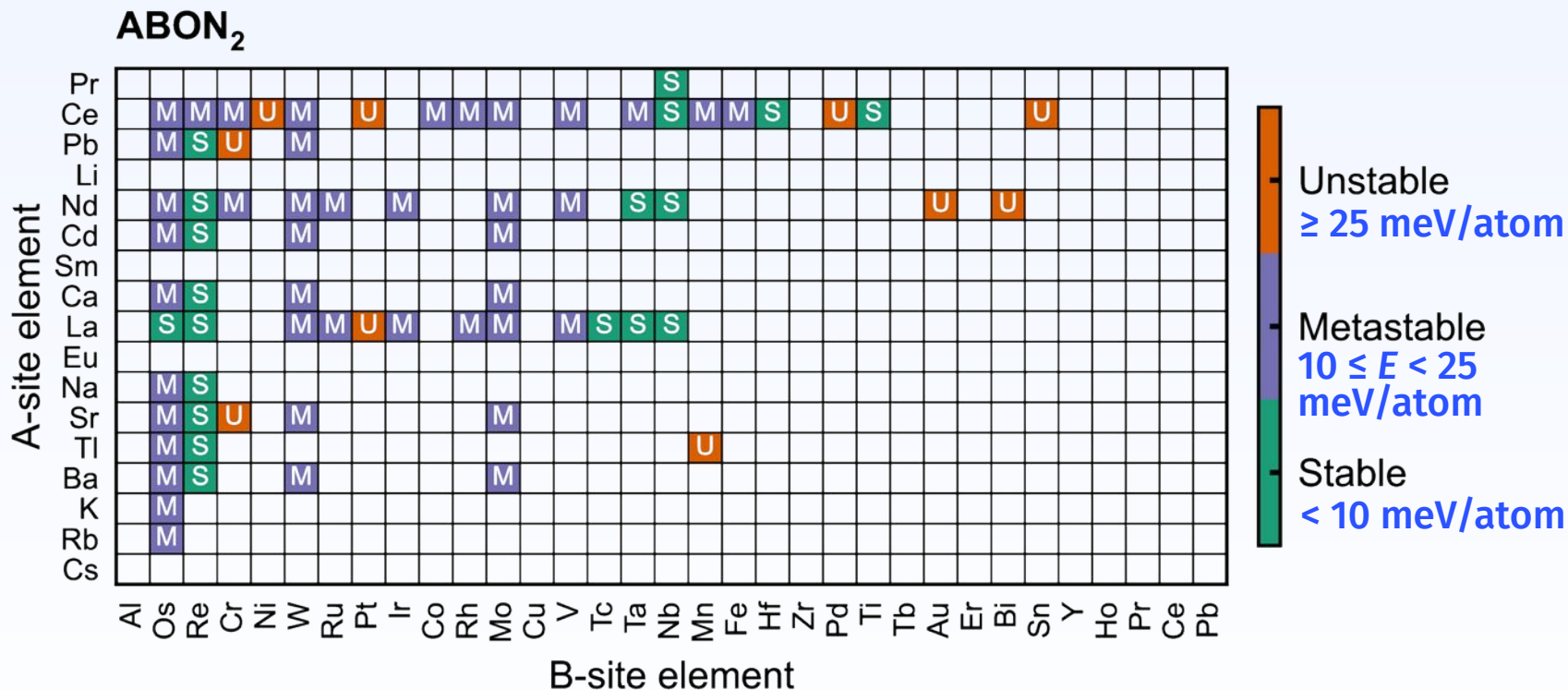
1. Wang, H.-C., Schmidt, J., Botti, S. & L. Marques, M. A. A high-throughput study of oxynitride, oxyfluoride and nitrofluoride perovskites. *Journal of Materials Chemistry A* **9**, 8501–8513 (2021).

We identify 66 stable ABO₂N PON materials



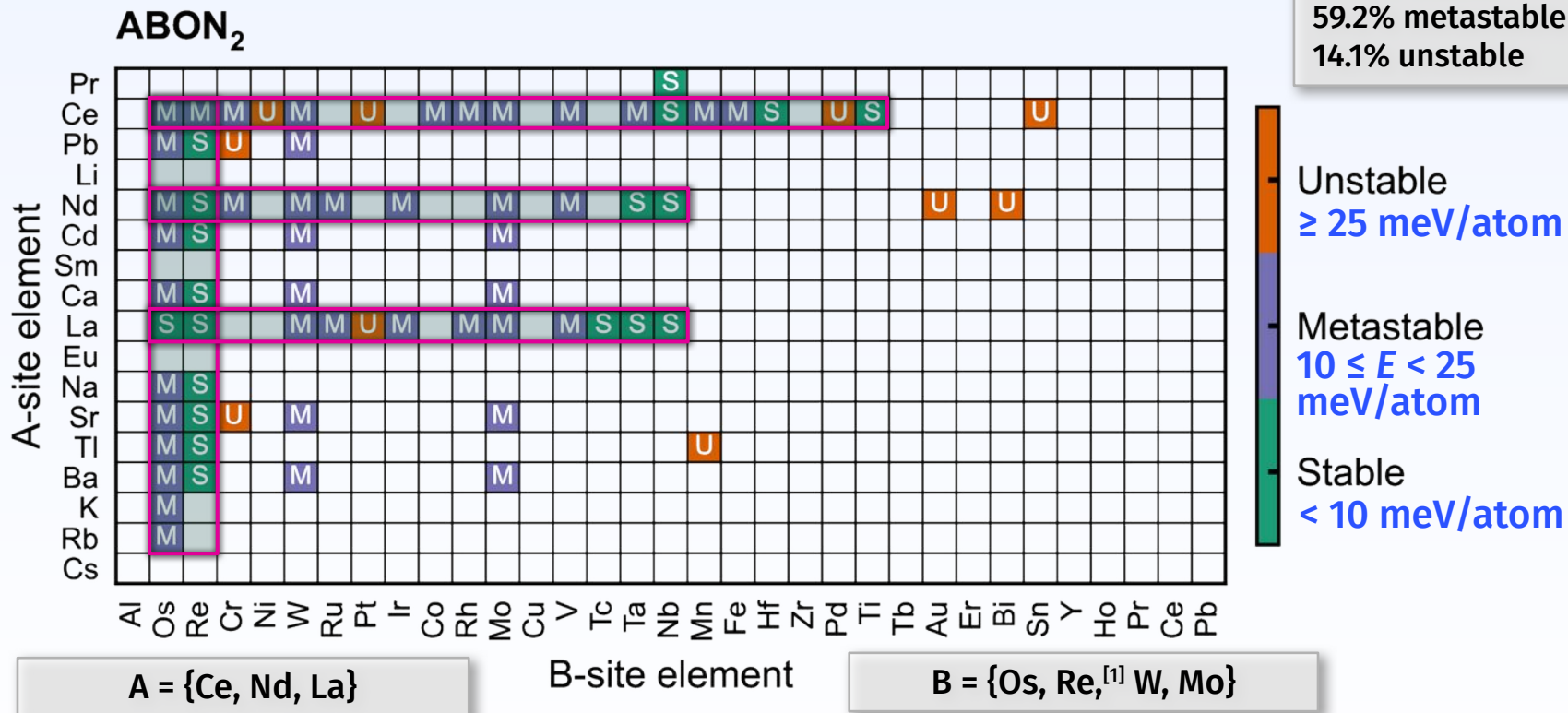
1. Wang, H.-C., Schmidt, J., Botti, S. & L. Marques, M. A. A high-throughput study of oxynitride, oxyfluoride and nitrofluoride perovskites. *Journal of Materials Chemistry A* **9**, 8501–8513 (2021).

We identify 19 stable ABO_2N PON materials



1. Wang, H.-C., Schmidt, J., Botti, S. & L. Marques, M. A. A high-throughput study of oxynitride, oxyfluoride and nitrofluoride perovskites. *Journal of Materials Chemistry A* **9**, 8501–8513 (2021).

We identify 19 stable ABO₂N PON materials

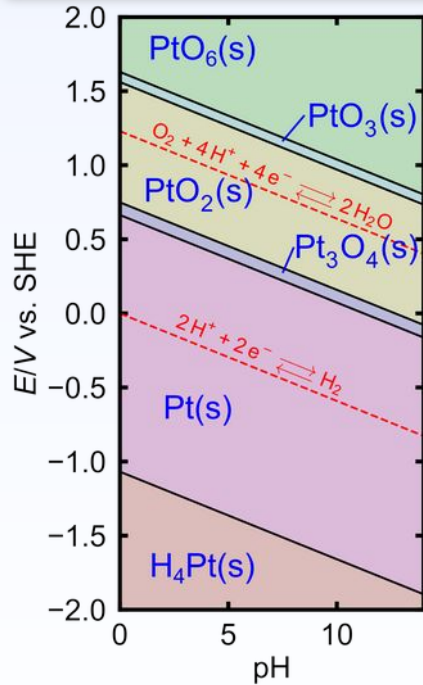


1. Wang, H.-C., Schmidt, J., Botti, S. & L. Marques, M. A. A high-throughput study of oxynitride, oxyfluoride and nitrofluoride perovskites. *Journal of Materials Chemistry A* **9**, 8501–8513 (2021).

We generate a Pourbaix diagram for LaTaO_2N

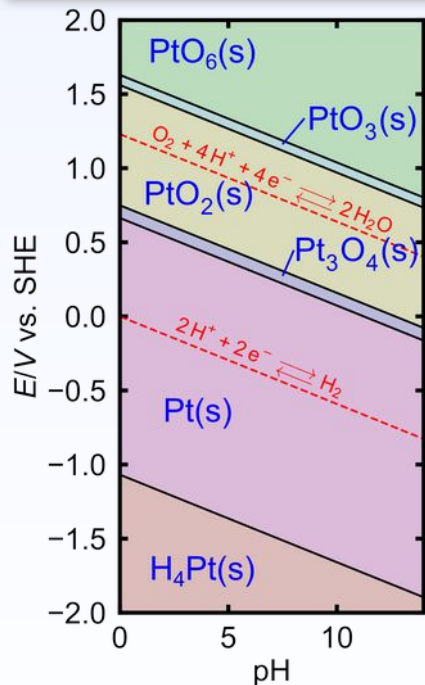
We generate a Pourbaix diagram for LaTaO₂N

Pourbaix diagram, Pt-O-H system

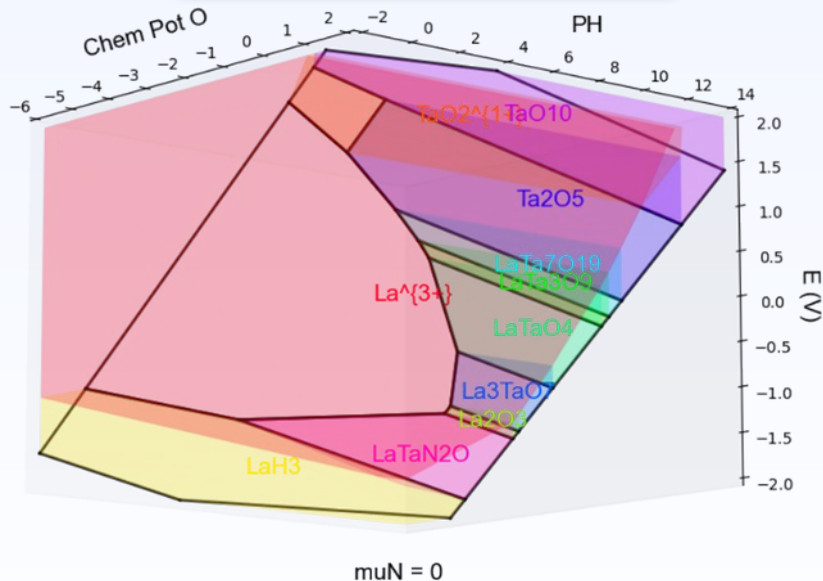


We generate a Pourbaix diagram for LaTaO₂N

Pourbaix diagram, Pt-O-H system



Pourbaix diagram, La-Ta-O-N-H system, $\mu_{\text{La}} = \mu_{\text{Ta}} = 0$



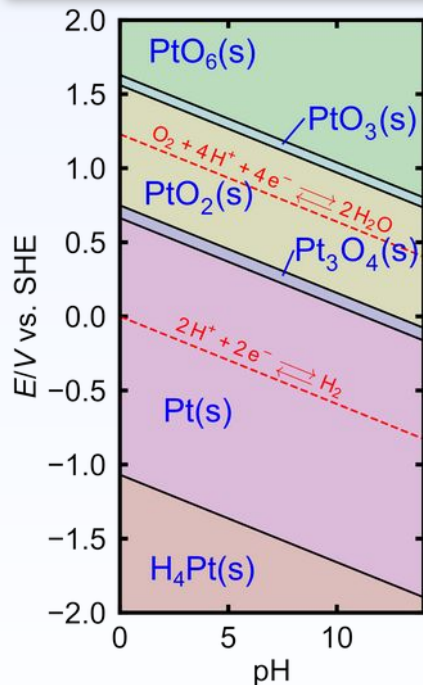
Animation



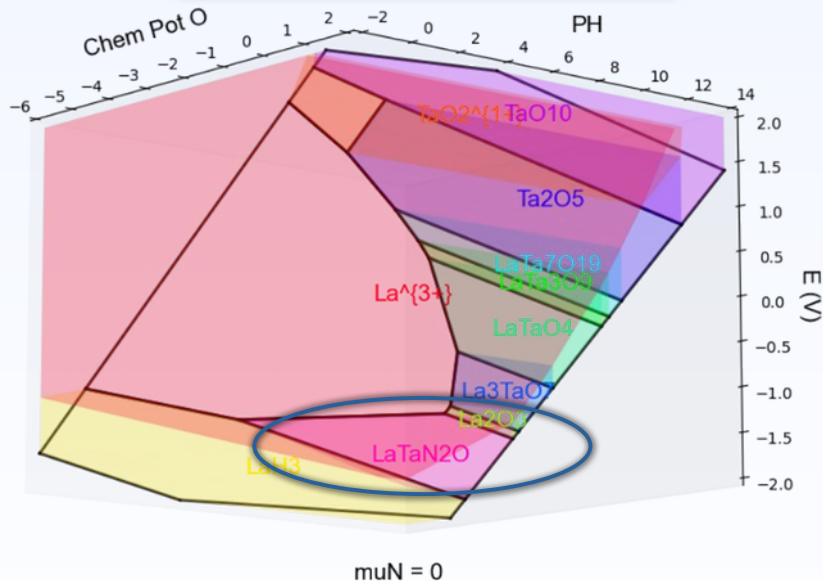
Jiadong Chen
Sun Research Group
Materials Science
University of Michigan

We generate a Pourbaix diagram for LaTaO₂N

Pourbaix diagram, Pt-O-H system



Pourbaix diagram, La-Ta-O-N-H system, $\mu_{La} = \mu_{Ta} = 0$



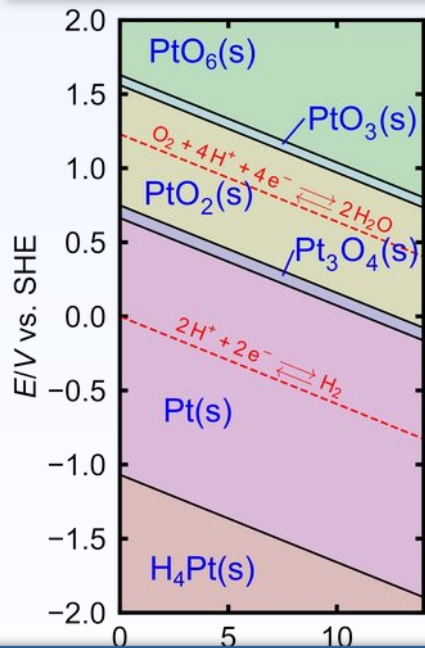
Animation



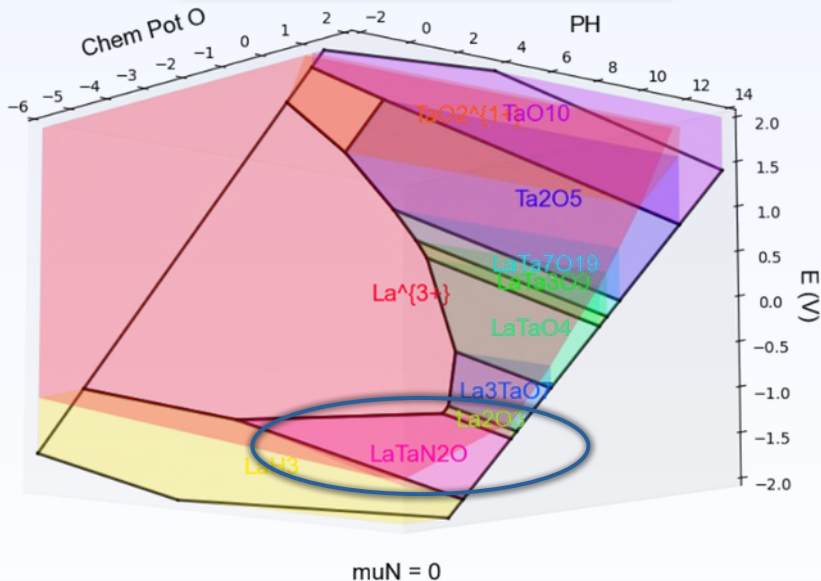
Jiadong Chen
Sun Research Group
Materials Science
University of Michigan

We generate a Pourbaix diagram for LaTaO₂N

Pourbaix diagram, Pt-O-H system



Pourbaix diagram, La-Ta-O-N-H system, $\mu_{\text{La}} = \mu_{\text{Ta}} = 0$



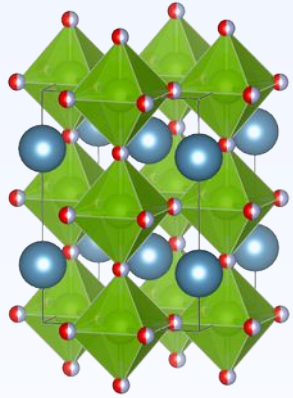
Animation



Jiadong Chen
Sun Research Group
Materials Science
University of Michigan

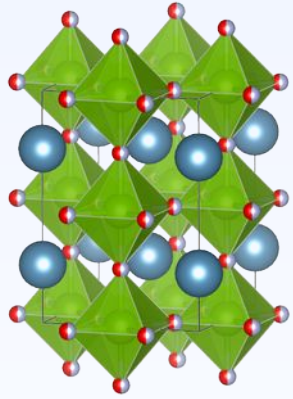
LaTaO₂N appears to be synthesizable with relative low NH₃, N₂ pressures and flowrates.

Found 85 stable PONs and that *cis* ordering is preferred in PON structures

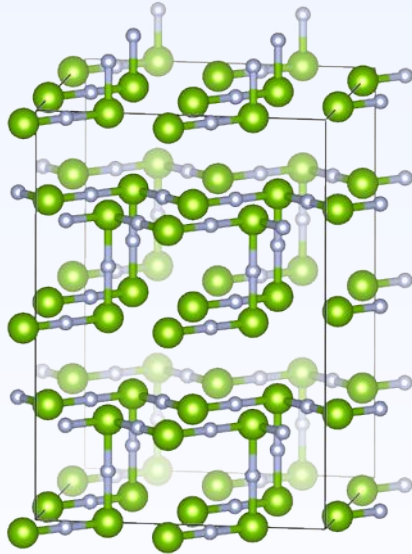


PON compound

Found 85 stable PONs and that *cis* ordering is preferred in PON structures

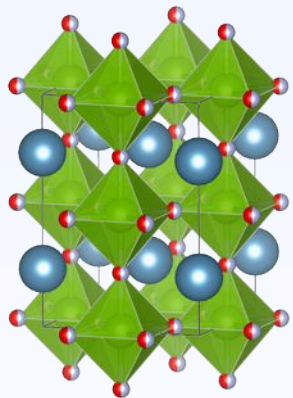


PON compound

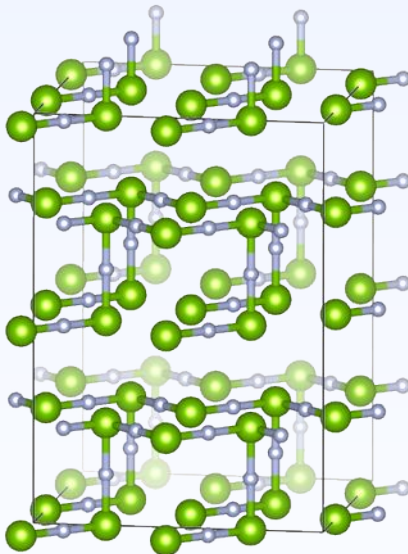


Anion ordering with
cis M—B—M bonds

Found 85 stable PONs and that *cis* ordering is preferred in PON structures

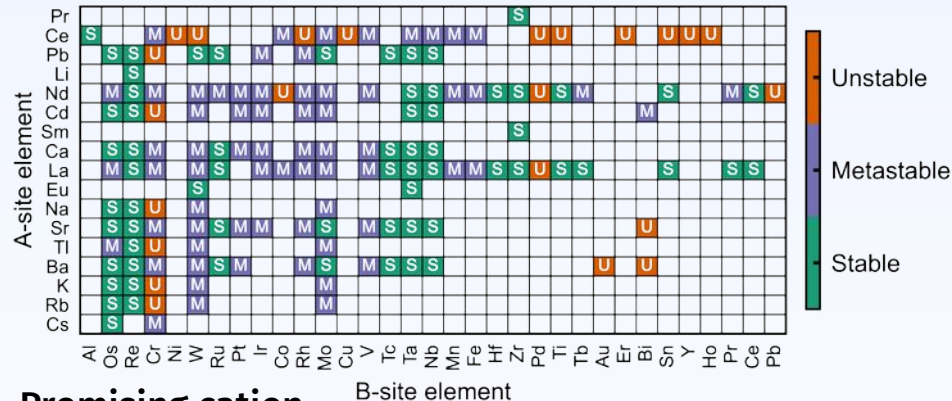


PON compound



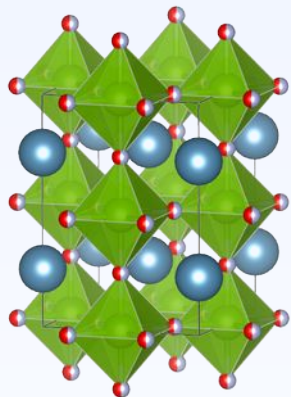
Anion ordering with *cis* M—B—M bonds

ABO₂N

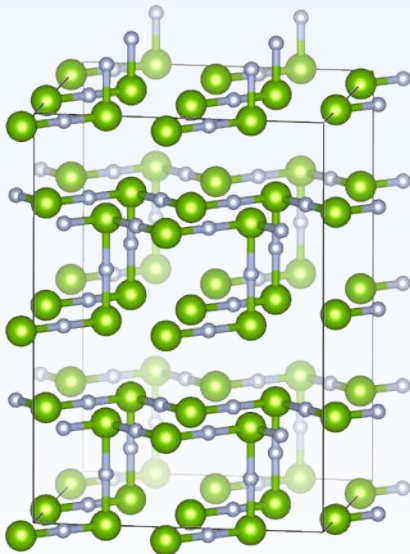


Promising cation chemistries

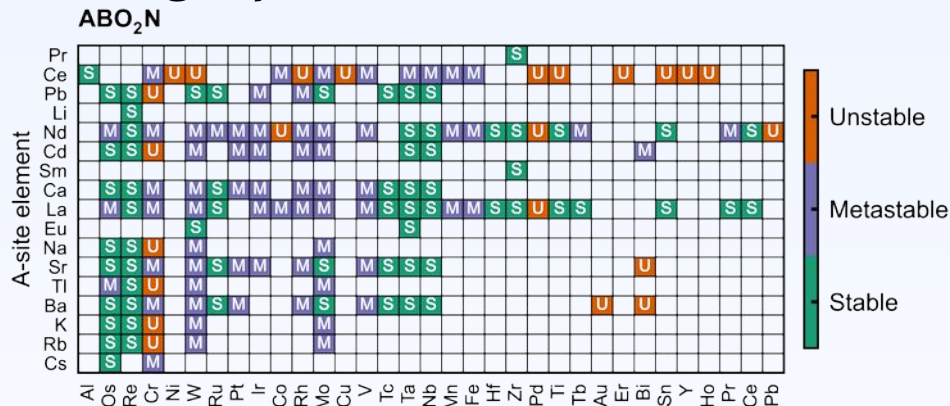
Found 85 stable PONs and that *cis* ordering is preferred in PON structures



PON compound

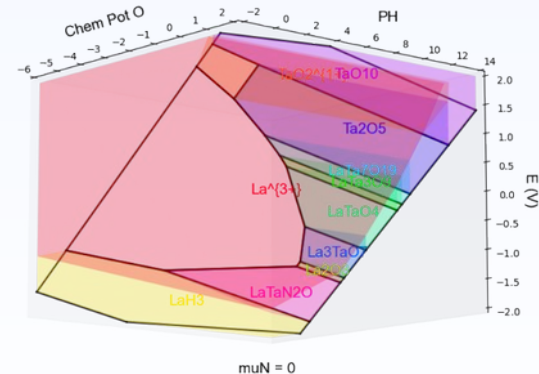


Anion ordering with *cis* M—B—M bonds

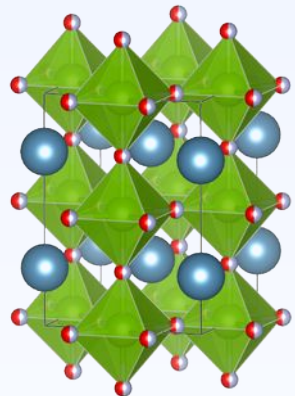


Promising cation chemistries

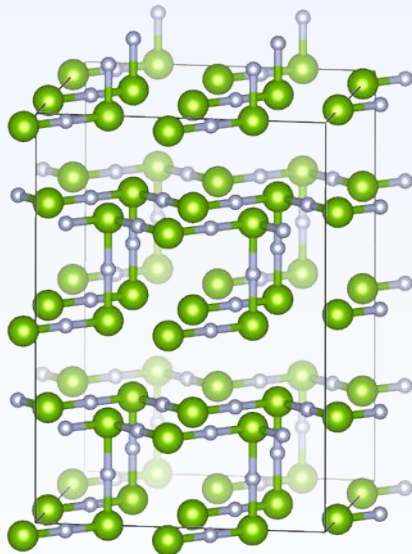
Developing electrochemical stability screening



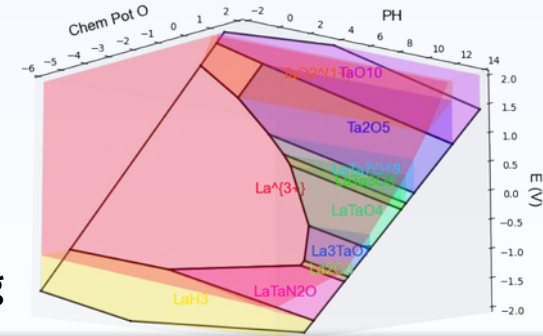
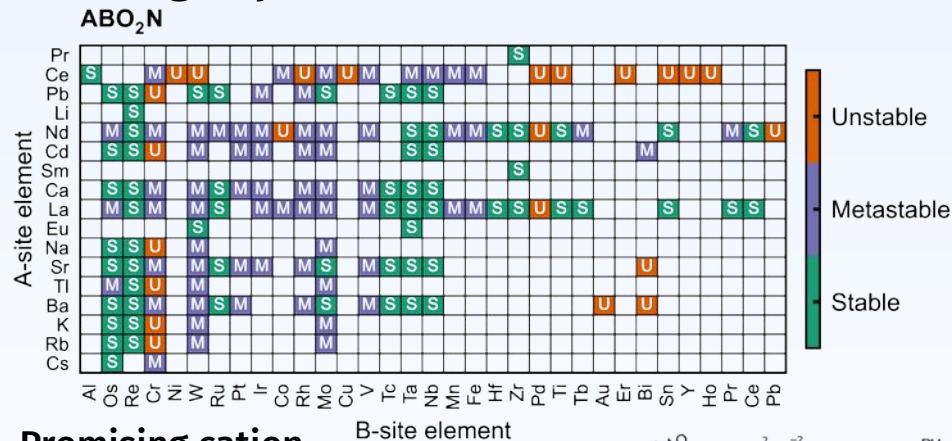
Found 85 stable PONs and that *cis* ordering is preferred in PON structures



PON compound



Anion ordering with *cis* M-B-M bonds



Next steps: synthesis!! Collaborating with LANL experimentalists to make stable candidates.



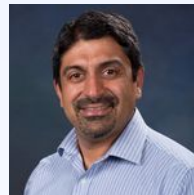
Acknowledgments



Bryan Goldsmith
Chemical Engineering
University of Michigan



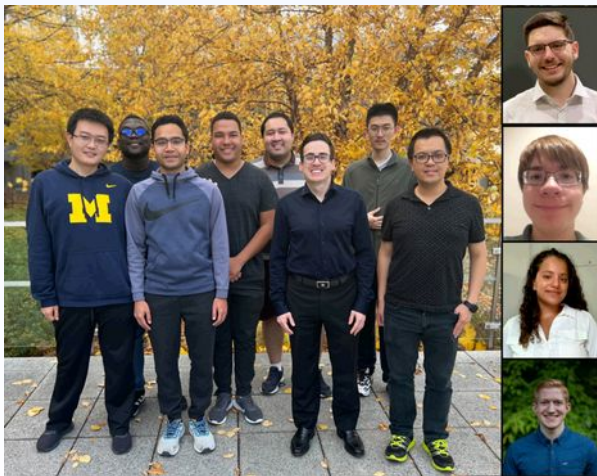
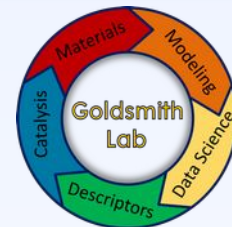
Bianca Ceballos
Materials Physics and Applications
Los Alamos National Laboratory



Ranganchary Mukundan
Materials Physics and Applications
Los Alamos National Laboratory



Jiadong Chen
Materials Science
University of Michigan



Amitava Banerjee
Metallurgical & Materials Engineering
IIT-Jodhpur



Ghanshyam Pilania
Materials Science and Technology
Los Alamos National Laboratory



Wenhao Sun
Materials Science
University of Michigan



samuelyd.github.com
Read our recent
perspective!



J. Robert Beyster Computational
Innovation Graduate Fellows Program



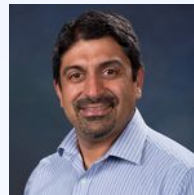
Acknowledgments



Bryan Goldsmith
Chemical Engineering
University of Michigan



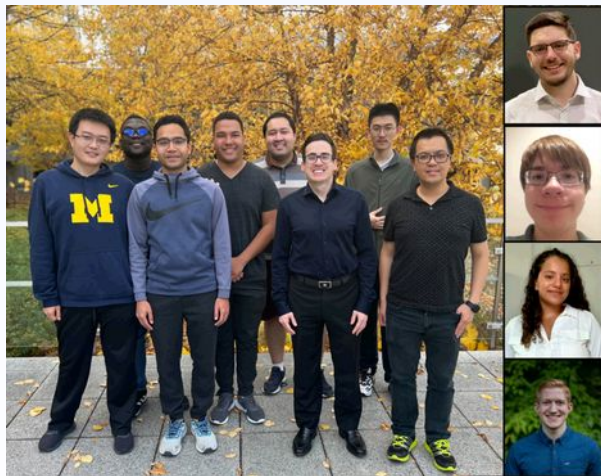
Bianca Ceballos
Materials Physics and Applications
Los Alamos National Laboratory



Ranganchary Mukundan
Materials Physics and Applications
Los Alamos National Laboratory



Jiadong Chen
Materials Science
University of Michigan



Amitava Banerjee
Metallurgical & Materials Engineering
IIT-Jodhpur

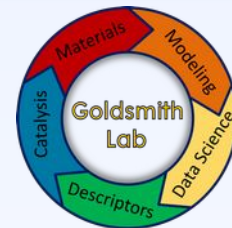
Ghanshyam Pilania
Materials Science and Technology
Los Alamos National Laboratory



Wenhao Sun
Materials Science
University of Michigan



J. Robert Beyster Computational
Innovation Graduate Fellows Program



samuelyd.github.com
Read our recent
perspective!



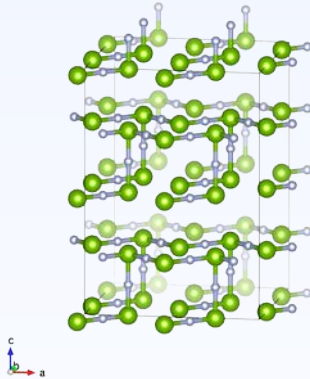
Questions?

Backup Slides

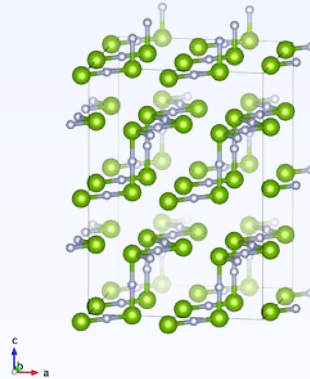
Low-energy orderings have a high degree of *cis* bonding

Low-energy orderings have a high degree of *cis* bonding

(B) (M)



ordering-0
100% global *cis* bonding
Cis counts: 1, 0, 3, 1
3/4 octahedra with *cis* bonds

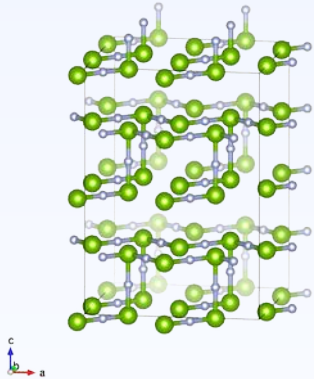


ordering-1
100% global *cis* bonding
Cis counts: 1, 0, 3, 1
3/4 octahedra with *cis* bonds

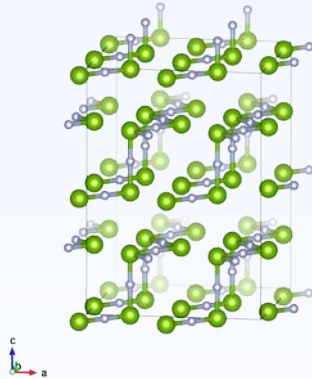


Low-energy orderings have a high degree of *cis* bonding

(B) (M)

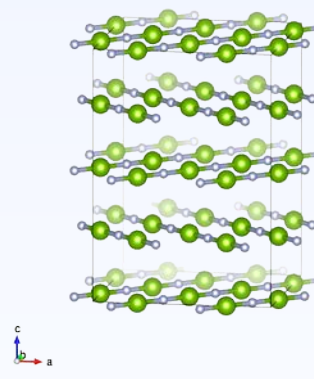


ordering-0
100% global *cis* bonding
Cis counts: 1, 0, 3, 1
3/4 octahedra with *cis* bonds

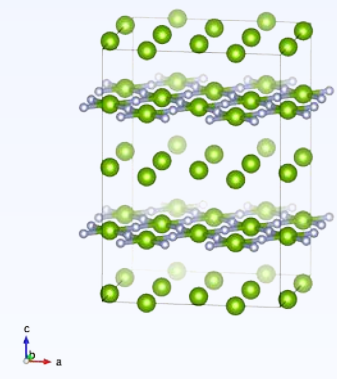


ordering-1
100% global *cis* bonding
Cis counts: 1, 0, 3, 1
3/4 octahedra with *cis* bonds

...



ordering-30
0% global *cis* bonding
Cis counts: 0, 0, 0, 0
0/4 octahedra with *cis* bonds



ordering-31
67% global *cis* bonding
Cis counts: 4, 0, 4, 0
2/4 octahedra with *cis* bonds



We aim to identify preferred anion orderings

- For $\sqrt{2} \times \sqrt{2} \times 2$ supercell, there are 32 total symmetrically distinct anion orderings.^[1]

We aim to identify preferred anion orderings

- For $\sqrt{2} \times \sqrt{2} \times 2$ supercell, there are 32 total symmetrically distinct anion orderings.^[1]

We aim to identify preferred anion orderings

- For $\sqrt{2} \times \sqrt{2} \times 2$ supercell, there are 32 total symmetrically distinct anion orderings.^[1]

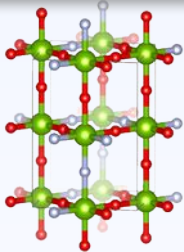
We aim to identify preferred anion orderings

- For $\sqrt{2} \times \sqrt{2} \times 2$ supercell, there are 32 total symmetrically distinct anion orderings.^[1]

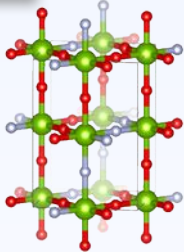
We aim to identify preferred anion orderings

- For $\sqrt{2} \times \sqrt{2} \times 2$ supercell, there are 32 total symmetrically distinct anion orderings.^[1]

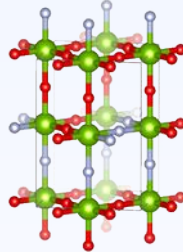
ABO₂N



ordering-0



ordering-1



ordering-2

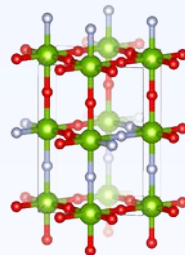
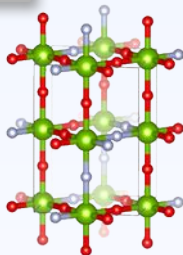
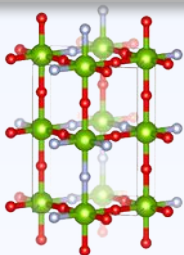
... (29 more)

1. Hart, G. L. W., Nelson, L. J. & Forcade, R. W. Generating derivative structures at a fixed concentration. *Computational Materials Science* **59**, 101–107 (2012).

We aim to identify preferred anion orderings

- For $\sqrt{2} \times \sqrt{2} \times 2$ supercell, there are 32 total symmetrically distinct anion orderings.^[1]

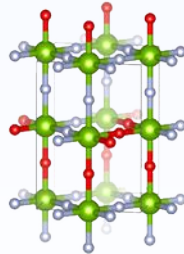
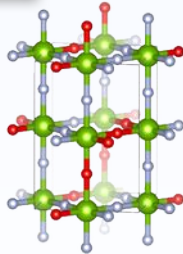
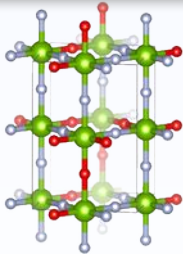
ABO₂N



... (29 more)

Swap O, N

ABON₂



1. Hart, G. L. W., Nelson, L. J. & Forcade, R. W. Generating derivative structures at a fixed concentration. *Computational Materials Science* **59**, 101–107 (2012).

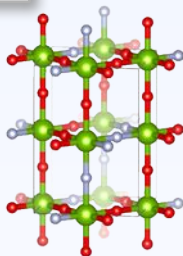
We aim to identify preferred anion orderings

- For $\sqrt{2} \times \sqrt{2} \times 2$ supercell, there are 32 total symmetrically distinct anion orderings.^[1]

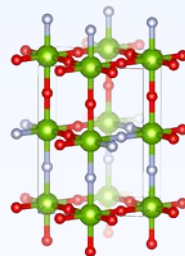
ABO₂N



ordering-0



ordering-1

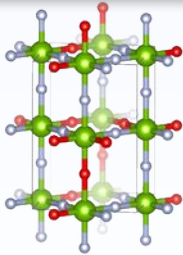


ordering-2

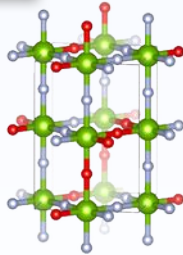
... (29 more)

Swap O, N

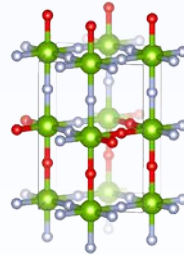
ABON₂



ordering-0



ordering-1



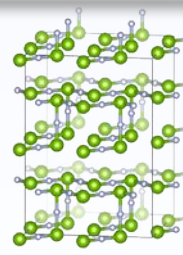
ordering-2

Look at M-B-M bonds only

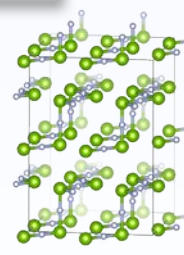
Topology



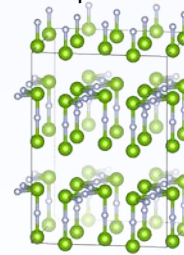
M = minority composition anion



ordering-0



ordering-1

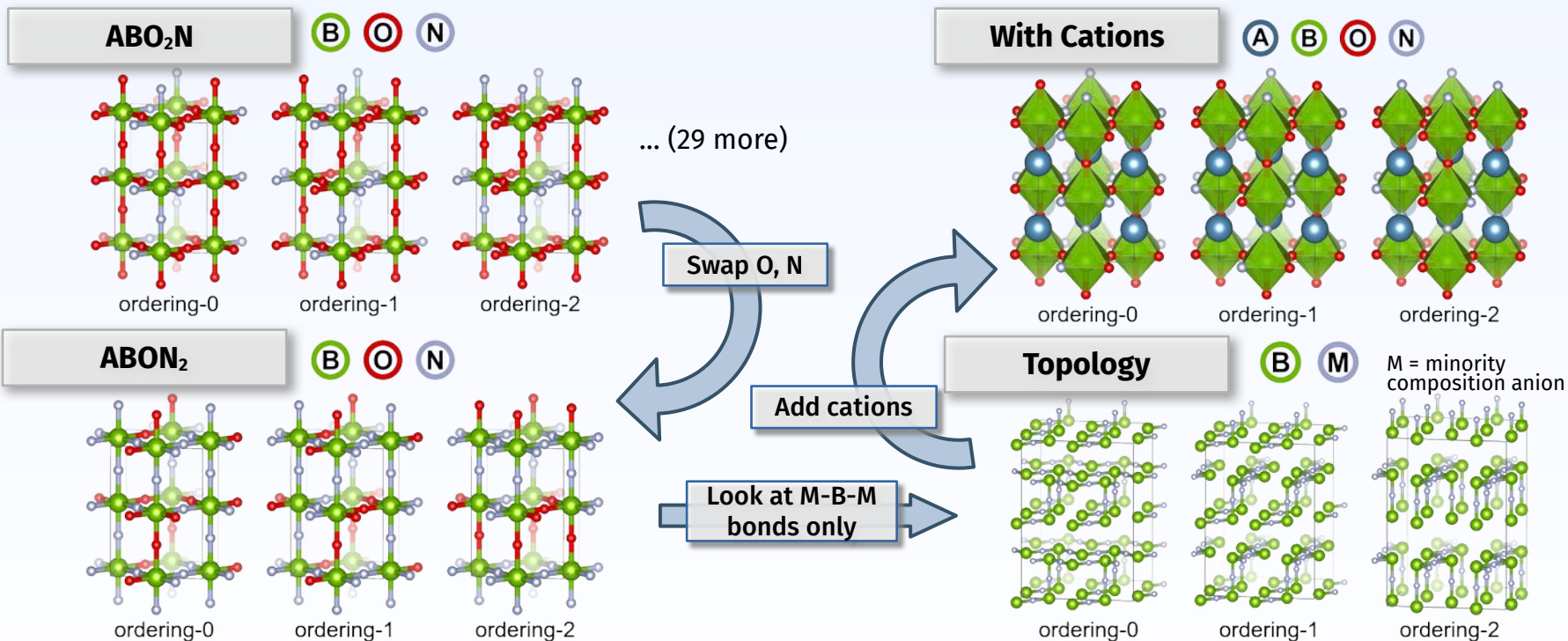


ordering-2

1. Hart, G. L. W., Nelson, L. J. & Forcade, R. W. Generating derivative structures at a fixed concentration. *Computational Materials Science* **59**, 101–107 (2012).

We aim to identify preferred anion orderings

- For $\sqrt{2} \times \sqrt{2} \times 2$ supercell, there are 32 total symmetrically distinct anion orderings.^[1]



1. Hart, G. L. W., Nelson, L. J. & Forcade, R. W. Generating derivative structures at a fixed concentration. *Computational Materials Science* **59**, 101–107 (2012).

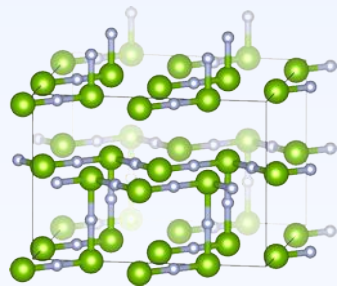
We screen 295 PON compounds and group by stability above convex hull

We screen 295 PON compounds and group by stability above convex hull



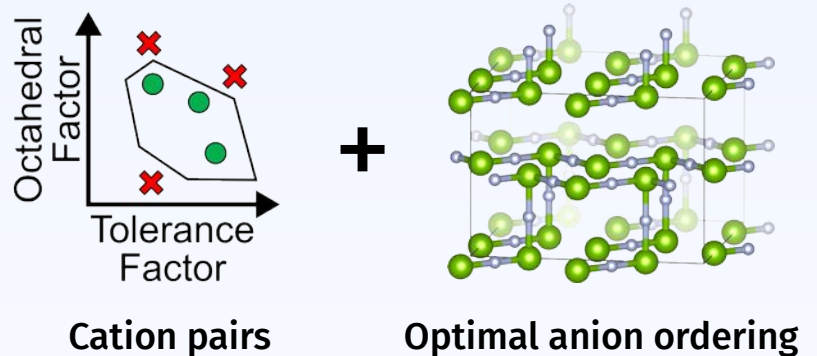
Cation pairs

+



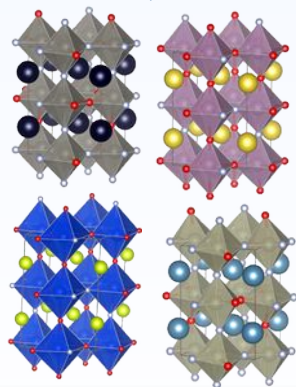
Optimal anion ordering

We screen 295 PON compounds and group by stability above convex hull



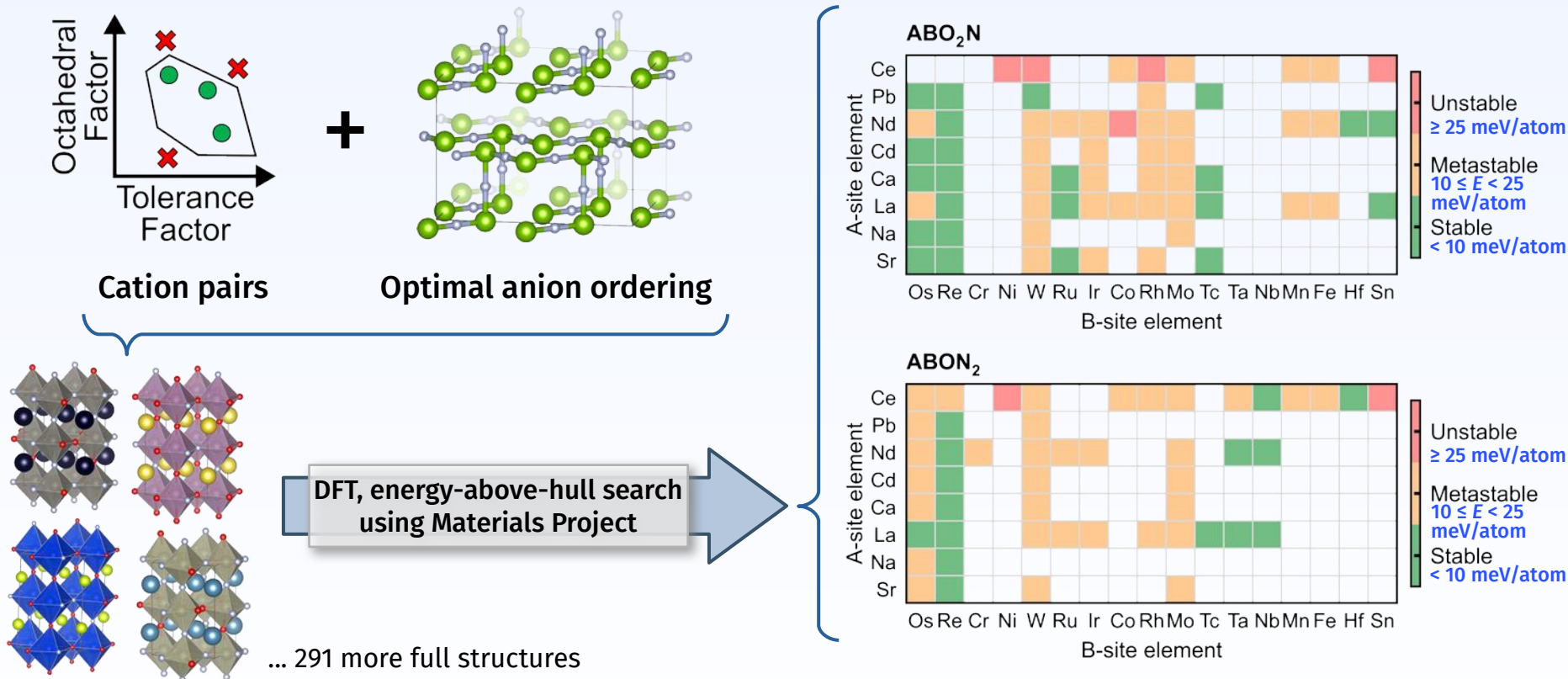
Cation pairs

Optimal anion ordering



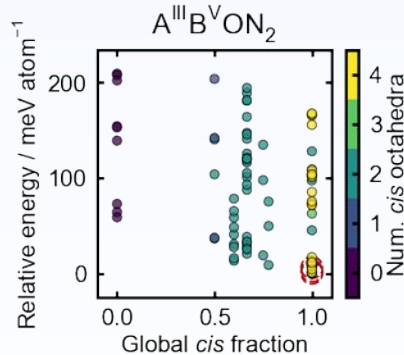
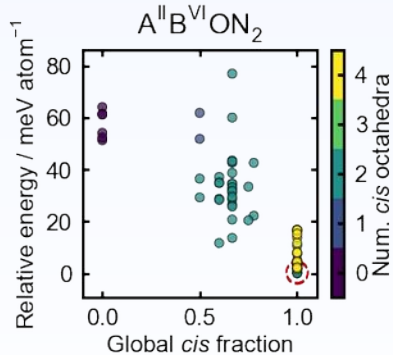
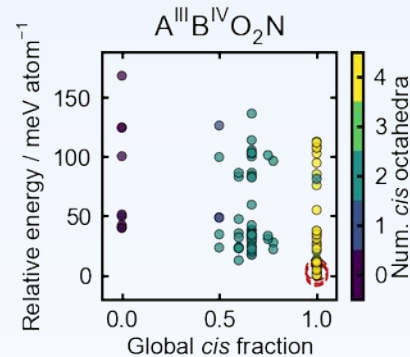
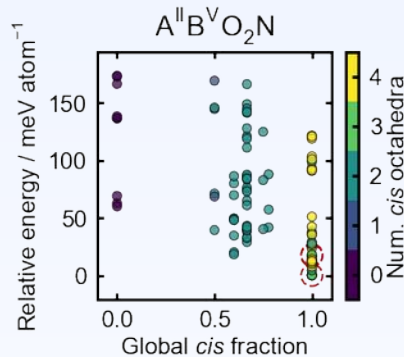
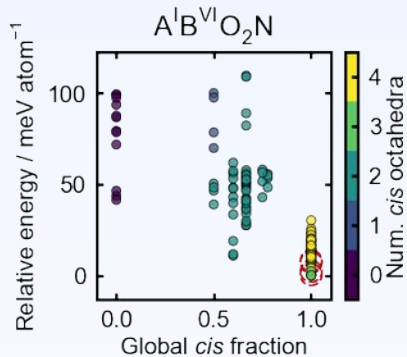
... 291 more full structures

We screen 295 PON compounds and group by stability above convex hull

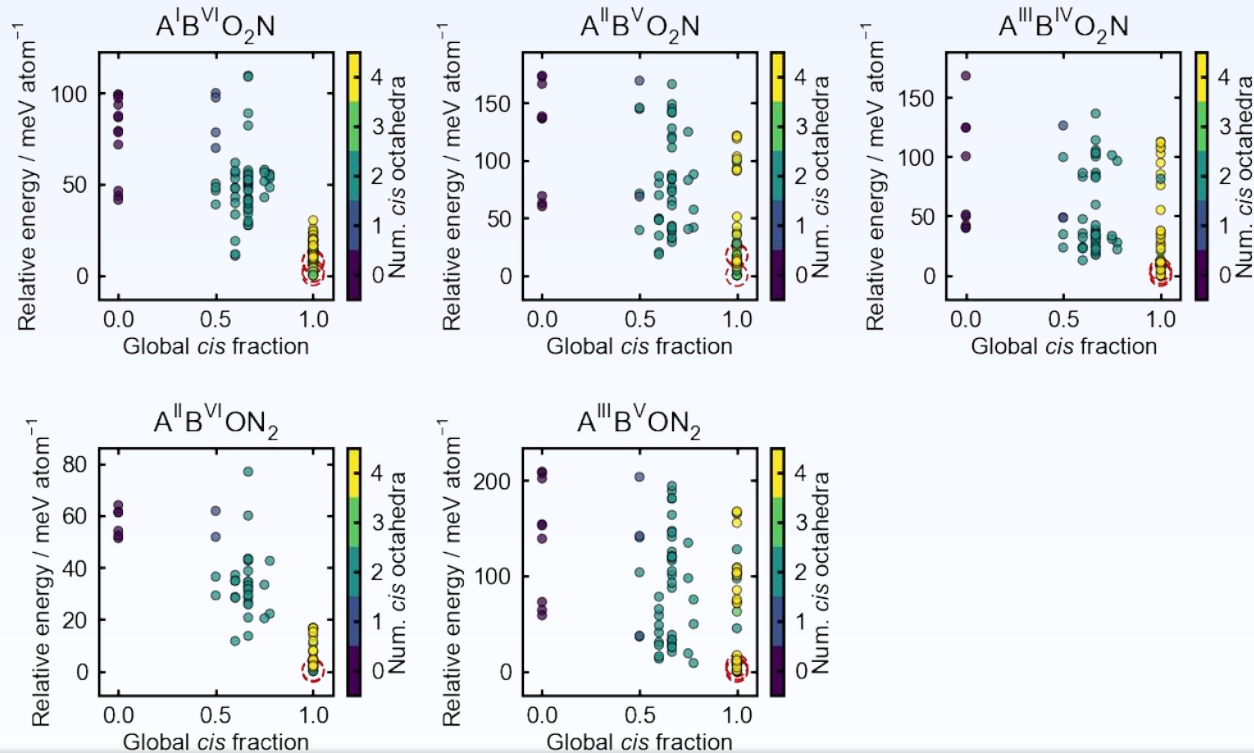


1. Jain, A. et al. Commentary: The Materials Project: A materials genome approach to accelerating materials innovation. *APL Materials* **1**, 011002 (2013).

A global *cis* fraction of 1 leads to the most stable anion ordering, for all cation pairs



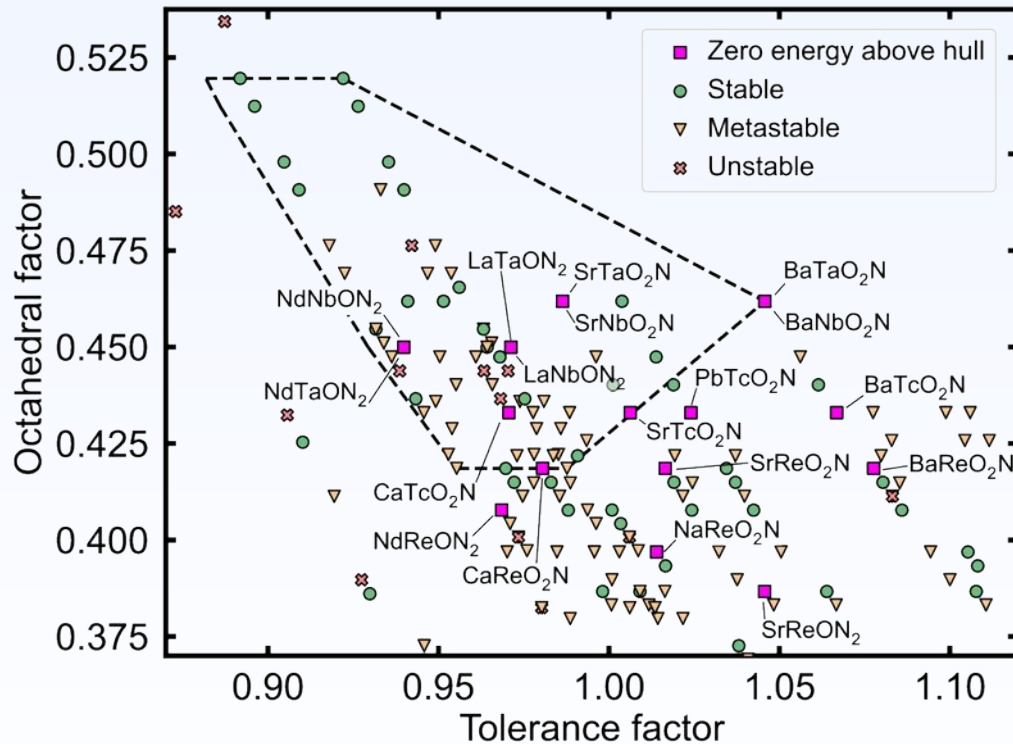
A global *cis* fraction of 1 leads to the most stable anion ordering, for all cation pairs



Correlations not strong across all cation pairs, but high fraction of global *cis* ordering is important.

DFT-predicted hull identifies new possible stable PON compounds for exploration

- B = Re compounds
- A = La, Ca, Pb compounds
- Many stable compounds are outside southeast border of experimental stability hull.



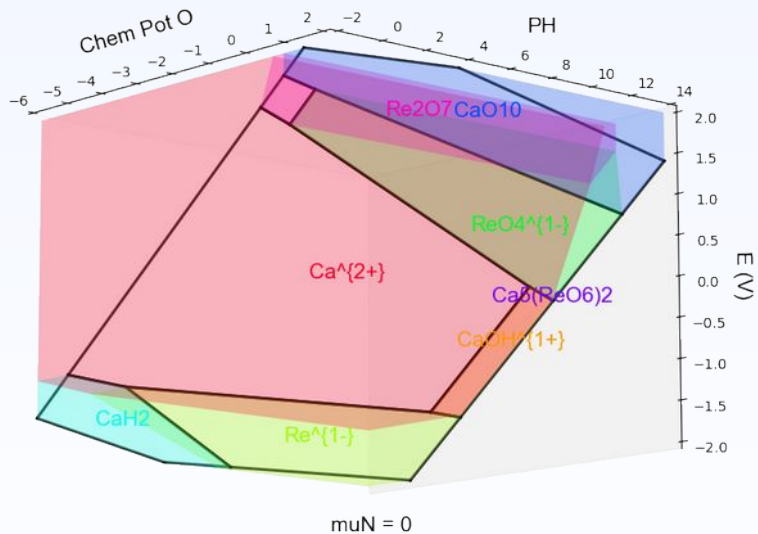
We generate a Pourbaix diagram for CaReO_2N



Jiadong Chen
Sun Research Group
Materials Science
University of Michigan

We generate a Pourbaix diagram for CaReO_2N

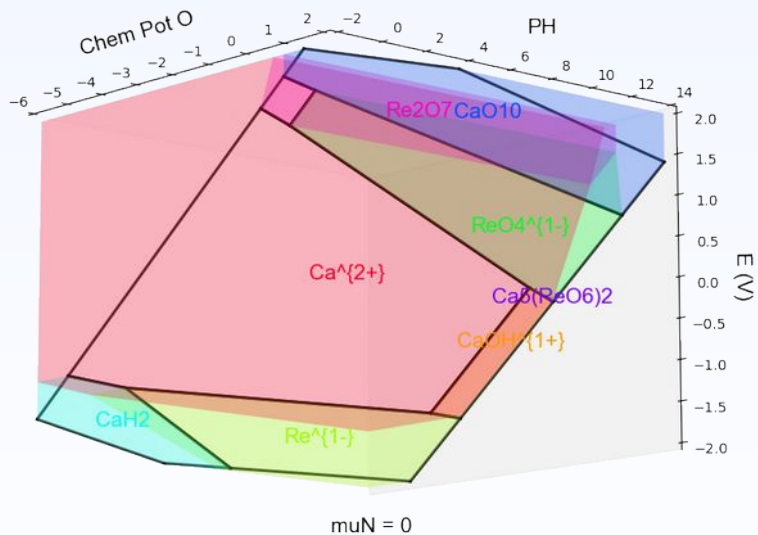
Pourbaix diagram, $\mu_{\text{Ca}} = \mu_{\text{Re}} = 0$



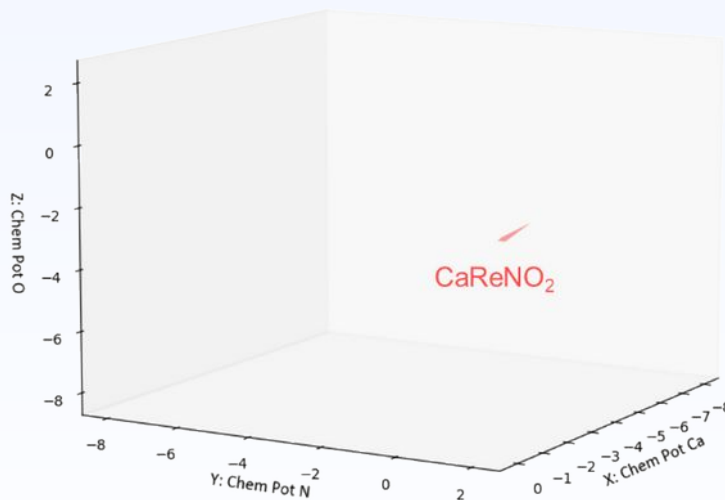
Jiadong Chen
Sun Research Group
Materials Science
University of Michigan

We generate a Pourbaix diagram for CaReO_2N

Pourbaix diagram, $\mu_{\text{Ca}} = \mu_{\text{Re}} = 0$



Stability region for solid PON

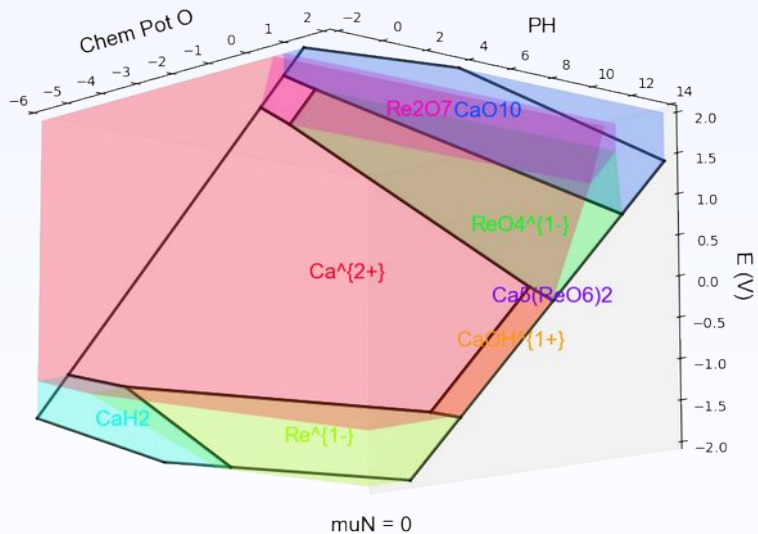


Jiadong Chen
Sun Research Group
Materials Science
University of Michigan

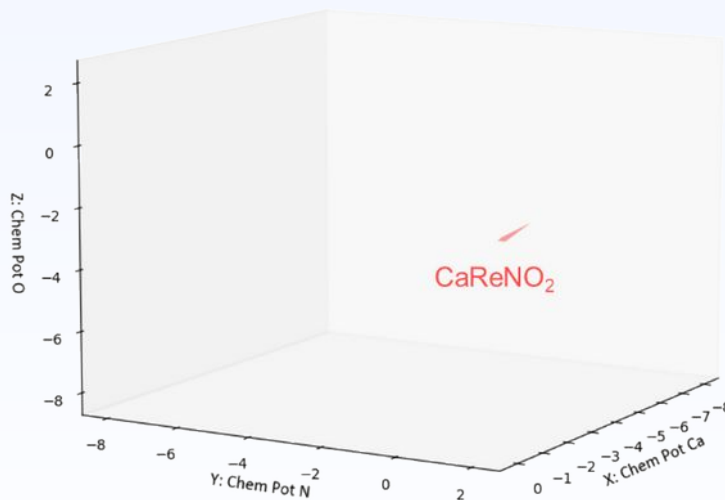
We generate a Pourbaix diagram for CaReO_2N

Pourbaix diagram, $\mu_{\text{Ca}} = \mu_{\text{Re}} = 0$

Animation



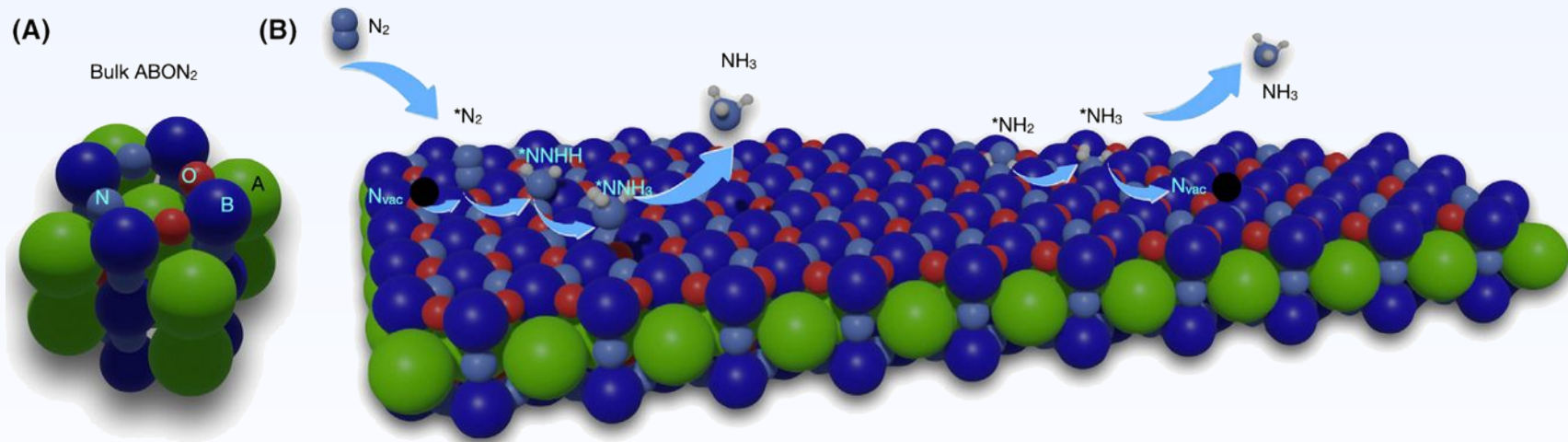
Stability region for solid PON



Jiadong Chen
Sun Research Group
Materials Science
University of Michigan

Synthesis could require very high partial pressures of NH_3 or N_2 precursor.

MvK mechanism for perovskite oxynitrides



Trends in Chemistry

Figure 1. Overview of the electrocatalytic nitrogen reduction reaction (ENRR) on an ABON₂ perovskite oxynitride. (A) Bulk unit cell of an ABON₂ perovskite oxynitride. (B) Illustration of an associative distal Mars-van-Krevelen mechanism, with N vacancies (N_{vac}, ●) facilitating associative adsorption and stepwise hydrogenation of N₂ to NH₃. Atom key: light blue = N, red = O, green = A, dark blue = B, gray = H.

Young, S. D., Banerjee, A., Paliana, G. & Goldsmith, B. R. Perovskite oxynitrides as tunable materials for electrocatalytic nitrogen reduction to ammonia. *Trends in Chemistry* **3**, 694–696 (2021).